

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

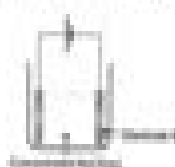
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
-----------------	--	-----------	-----	----------------

4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV.



4.8 The diagram shows the setup for reaction IV. The reaction is carried out in a beaker. The gas is collected in a test tube. The setup is labeled 'Reaction IV'.



4.9 Identify the reagent used in reaction I.

Reaction I: $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Br} + \text{NaOH} \rightarrow \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH} + \text{NaBr}$

4.10 The diagram shows the setup for reaction I. The reaction is carried out in a beaker. The gas is collected in a test tube. The setup is labeled 'Reaction I'.

4.11 The diagram shows the setup for reaction I. The reaction is carried out in a beaker. The gas is collected in a test tube. The setup is labeled 'Reaction I'.

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