



A) atoms only

B) mass only

C) mass and atoms only

D) moles only

9. What is the limiting reactant when 5 moles of H_2 react with 3 moles of N_2 to form NH_3 ?

A) H_2

B) N_2

C) NH_3

D) None

10. The relationship between actual yield and the theoretical yield is that they are

A) always equal

B) often equal

C) actual is larger than theoretical

D) theoretical is larger

Short Answer

1. Calculate the mass in grams, of a sample of $\text{KHC}_2\text{H}_3\text{O}_6$, that contains 6.54×10^{24} atoms of hydrogen. (4T, 3C)

2. In our diets we consume a lot of fat. A fat is composed, in part, of long chains of carbon and hydrogen atoms. You decide to analyze some leftover turkey fat from your Thanksgiving dinner and find it contains 64.67% C, 9.76% H, and 25.5% O by mass. Calculate its empirical and molecular formula if the fat has a molar mass of 372.56 g/mol (4A, 1C)

Physical Science Chemistry Test Answers Unit

O García



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