#### TYESC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte?
	a) KCI
	b) C <sub>i</sub> Cl <sub>2</sub>
	d) AlCli
	d) MgCl <sub>2</sub>
2	The ionic strength of a solution of molality m, and valency of the ion Zi is
	a) $\mu = \frac{\epsilon}{2} m_i Z \ell^2$
	b) $\mu = \sum_{i=1}^{n} m_i Z i^2$
	$\epsilon \lambda = \frac{1}{2} \sum Z \ell^2$
	d) $\mu = \frac{1}{2} \sum m_i Z \ell^2$
3	Which of the following is an example of electrolyte concentration cell with transference?
	ii) Cii [CuSO <sub>1</sub> ] [AgNO <sub>1</sub> ] Ag
	b) Pt. H <sub>2g</sub>   HCl <sub>sq</sub>   AgCl <sub>30</sub> Ag
	c) Pt. H <sub>20</sub>   HCl <sub>m</sub> (a)   AgNO <sub>2</sub> (a)   Ag
	d) Ag – AgCl [NaCl(a <sub>0</sub> )] NaCl(a <sub>0</sub> )[AgCl - Ag
4	A concentration cell which has salt bridge in its representation is
	electrolyte concentration cell with transference
	b) Electrode concentration cell
	c) Electrolyte concentration cell without transference
	d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by
	a) $E_j = E_{exidation} + E_{reduction}$
	b) $E_{cell} = E_{saldation} + E_{reduction} + E_{j}$
	c) $E_{cett} = E_{exidaction} - E_{reduction} - E_{j}$
	d) $E_j = E_{oxidation} - E_{reduction}$
6	The activity 'a' is related to the activity coefficient " y' by the equation
	a) a = = =
	b) $y = a - m$
	c) $\alpha = m + \gamma$
	d) $a = my$

# **Physical Chemistry Multiple Choice**

S. K. Bhasin

## **Physical Chemistry Multiple Choice:**

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