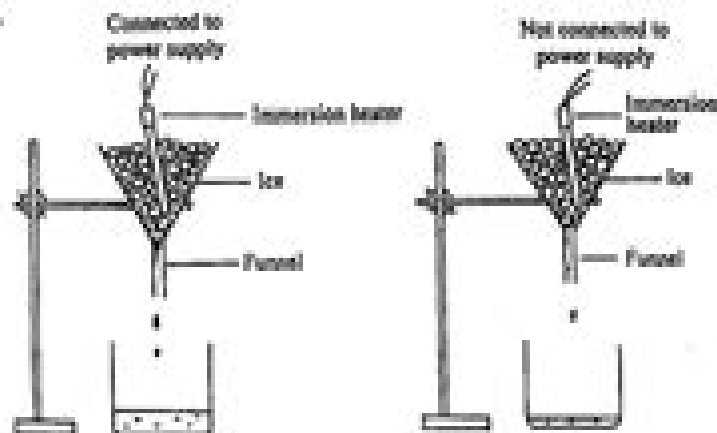


19. Which of the following statements about internal energy, heat and temperature is/are true?

- (1) The internal energy of a body is a measure of the total kinetic energy and potential energy of the molecules in the body.
- (2) Two bodies of the same temperature always have the same amount of internal energy.
- (3) Heat is a measure of the energy transferred from one body to another as a result of temperature difference between the two bodies.

- A. (1) only
- B. (2) only
- C. (1) and (3) only
- D. (2) and (3) only
- E. (1), (2) and (3)

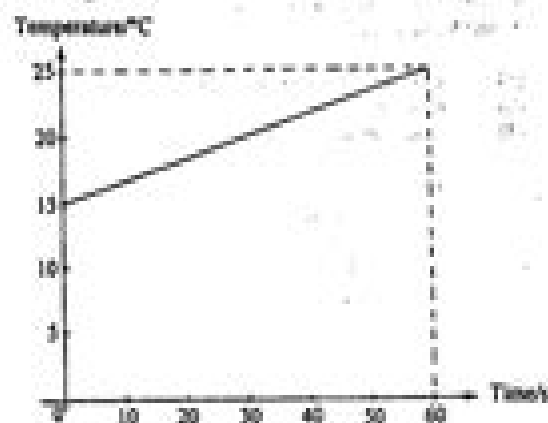
20.



The above set-up can be used to find the specific latent heat of fusion of ice. Which of the following is not an essential precaution to ensure an accurate result?

- A. using crushed ice in the experiment
- B. using melting ice in the experiment
- C. inserting the immersion heater into the ice completely
- D. using the same amount of ice in both funnels
- E. covering the funnels with lids

21.



The above graph shows the variation of the temperature of a liquid with time when the liquid is heated by a 400 W heater. The mass of the liquid is 2 kg. Find the specific heat capacity of the liquid. Assume all the energy given out by the heater is absorbed by the liquid.

- A. $83 \text{ J kg}^{-1} \text{ K}^{-1}$
- B. $480 \text{ J kg}^{-1} \text{ K}^{-1}$
- C. $1\,200 \text{ J kg}^{-1} \text{ K}^{-1}$
- D. $2\,400 \text{ J kg}^{-1} \text{ K}^{-1}$
- E. $12\,000 \text{ J kg}^{-1} \text{ K}^{-1}$

22.

A block of melting ice with mass 0.02 kg is put into a polystyrene cup containing 0.3 kg of water with initial temperature 20°C . After the mixture is stirred well, the ice block melts completely and the final temperature of the water becomes 14°C . Which of the following equations can be used to find the specific latent heat of fusion of ice, L ?

(Given : Specific heat capacity of water = $4200 \text{ J kg}^{-1} \text{ K}^{-1}$)

- A. $0.3 \times 4200 \times 6 = 0.02 L$
- B. $0.3 \times 4200 \times 6 = 0.02 L + 0.02 \times 4200 \times 6$
- C. $0.3 \times 4200 \times 6 = 0.02 L + 0.02 \times 4200 \times 6$
- D. $0.3 \times 4200 \times 6 = (0.02 L + 0.02 \times 4200) \times 14$
- E. $0.3 \times 4200 \times 6 = 0.02 L + 0.02 \times 4200 \times 14$

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Michael Brown



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