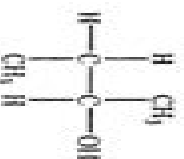


16. Which of the following conversions is NOT a reduction?

- A. $\text{Fe}_2\text{O}_3 \rightarrow \text{Fe}$ ✓
 B. $\text{Ca(OH)}_2 \rightarrow \text{CaO}$ ✓
 C. $\text{CH}_3\text{CO}_2\text{H} \rightarrow \text{CH}_3\text{CH}_2\text{OH}$
 D. $\text{H}_2\text{SO}_4 \rightarrow \text{SO}_2$

17. An organic compound has the following structure:



The systematic name of this compound is

- A. 1,2-dimethyl ethanol
 B. 1-methylpropan-1-ol
 C. 1-methylpropan-2-ol
 D. butan-2-ol

18. Consider the following information about three elements, X, Y and Z.

Element	Atomic number
X	12 (2, 8)
Y	16 (2, 8, 6)
Z	17 (2, 8, 7)

Which of the following statements concerning X, Y and Z is correct?

- A. X reacts with Z to form an ionic compound.
 B. Y is a stronger oxidising agent than Z.
 C. X has a simple molecular structure.
 D. Y can conduct electricity in the molten state.

19. Soap was prepared by heating fat with sodium hydroxide solution for some time. Concentrated sodium chloride solution was then added to the resulting mixture. The purpose of adding concentrated sodium chloride solution is

- A. to help the precipitation of the soap.
 B. to enhance the cleansing power of the soap.
 C. to reduce the alkalinity of the soap.
 D. to act as a preservative for the soap.

20.

A sample of zinc granules of mass 1.5 g was added to 100 cm³ of 0.25 M hydrochloric acid. What is the theoretical volume of hydrogen produced at room temperature and pressure?

(Relative atomic mass : Zn = 65.4;
 molar volume of gas at room temperature and pressure = 24 dm³)

- A. 0.30 dm³
 B. 0.33 dm³
 C. 0.60 dm³
 D. 0.60 dm³

21.

Which of the following solutions would NOT undergo a colour change when mixed with sodium sulphate solution?

- A. iodine solution
 B. acidified potassium permanganate solution
 C. chromium(III) sulphate solution
 D. iron(III) sulphate solution

22.

X is a bleaching agent. X bleaches a dye by oxidation and becomes Y at the end of the reaction. Which of the following combinations is correct?

X Y

- A. $\text{Cl}_2(\text{aq})$ $\text{Cl}(\text{aq})$
 B. $\text{Cl}_2(\text{aq})$ $\text{OCl}(\text{aq})$
 C. $\text{SO}_2(\text{aq})$ $\text{SO}_3^{2-}(\text{aq})$
 D. $\text{SO}_2(\text{aq})$ $\text{SO}_4^{2-}(\text{aq})$

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Vanessa Kind, Keith Taber



Paper 2 Chemistry 2003 Gr 11:

Overcoming Students' Misconceptions in Science Mageswary Karpudewan,Ahmad Nurulazam Md Zain,A.L. Chandrasegaran,2017-02-28 This book discusses the importance of identifying and addressing misconceptions for the successful teaching and learning of science across all levels of science education from elementary school to high school It suggests teaching approaches based on research data to address students common misconceptions Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included The science education literature extensively documents the findings of studies about students misconceptions or alternative conceptions about various science concepts Furthermore some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them In response this book offers an essential and easily accessible guide **35 JEE Main ONLINE & OFFLINE Physics, Chemistry & Mathematics Topic-wise Solved Papers - 4th Edition** Disha Experts,2017-08-31 The book 35 JEE Main Physics Chemistry Mathematics Online Offline Topic wise Solved Papers provides the last 16 years ONLINE OFFLINE 2002 17 papers The book contains a total of 35 papers 17 papers of AIEEE JEE Main from the year 2002 2017 held OFFLINE including the AIEEE 2011 RESCHEDULED paper and 18 JEE Main papers held ONLINE from 2012 17 The books are distributed into around 28 31 27 topics in Physics Chemistry Mathematics respectively exactly following the chapter sequence of the NCERT books of class 11 and 12 The questions in each topic are immediately followed by their detailed solutions The book constitutes around 4100 most important MCQs Children's Books in Print, 2007 ,2006 **Science** Vanessa Kind,Keith Taber,2005 This is a text for new teachers of science in the years of their early professional development including those on PGCE courses those in their induction year and those in years two and three of their teaching career **Index Medicus** ,2004 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **Handbook of Research on Science Education** Sandra K. Abell,Norman G. Lederman,2013-03-07 This state of the art research Handbook provides a comprehensive coherent current synthesis of the empirical and theoretical research concerning teaching and learning in science and lays down a foundation upon which future research can be built The contributors all leading experts in their research areas represent the international and gender diversity that exists in the science education research community As a whole the Handbook of Research on Science Education demonstrates that science education is alive and well and illustrates its vitality It is an essential resource for the entire science education community including veteran and emerging researchers university faculty graduate students practitioners in the schools and science education professionals outside of universities The National Association for Research in Science Teaching NARST endorses the Handbook of Research on Science Education as an important and valuable

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