

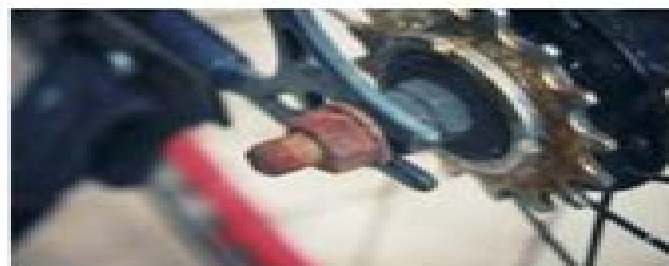
## Physical and Chemical Changes

Now that you know the difference between a physical and chemical property, this information can be used to determine whether a change in matter can be identified as a physical or chemical change. Just like the properties of matter, physical and chemical changes of matter have different characteristics which can usually be easily identified.

<u>Physical Changes</u>	<u>Chemical Changes</u>
<ul style="list-style-type: none"><li>• A change in size, shape, color, or a phase change (solid, liquid, gas)</li><li>• No new substance is formed and the molecules do not change</li></ul>	<ul style="list-style-type: none"><li>• A change in the physical and chemical properties</li><li>• A new substance is formed and the molecules of the substance are changed</li></ul>
<u>Examples of Physical Changes</u>	<u>Examples of Chemical Changes</u>
<ol style="list-style-type: none"><li>1. Melting a cube of sugar in coffee because it is still sugar</li><li>4. Water changed to an ice cube, boiling water, or a melting ice cube, molecules remain the same</li><li>5. Cutting apart an apple into smaller pieces making it easier to eat</li></ol>	<ol style="list-style-type: none"><li>1. Burning a cube of sugar with fire changes its molecules</li><li>4. Baking the ingredients of a cake changes many of the molecules due to the heat</li><li>5. An apple beginning to rot or being digested inside a person's stomach after it is eaten, will both change its molecules</li></ol>

Chemical and physical changes take place every day all around you. If you change the way something looks, but have not made a new substance, it is a **physical change**. If the substance has been changed into another substance it is a **chemical change**.

In a chemical change the atoms making up the molecules change. There is a **chemical reaction** that takes place during a chemical change. The reaction takes place usually due to another substance or heat added to it. One of the most commonly seen chemical reaction is rust forming on metal. If you leave your bike in the rain, it may begin to rust. Iron is exposed to oxygen gas in the air which results in rust. The iron is **oxidized** (oxygen added) and becomes iron oxide, which is the chemical name for rust.



There are many, many other examples of physical and chemical changes. In summary, if the molecules stay the same it is a physical change, if the molecules change, it is a chemical change.



**Name three of the same physical and chemical changes that take place in your life each day.**

*Learn*  
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# Physical And Chemical Changes Lessons

**Xiang Xie**



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## **Table of Contents Physical And Chemical Changes Lessons**

1. Understanding the eBook Physical And Chemical Changes Lessons
  - The Rise of Digital Reading Physical And Chemical Changes Lessons
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical And Chemical Changes Lessons
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical And Chemical Changes Lessons
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical And Chemical Changes Lessons
  - Personalized Recommendations
  - Physical And Chemical Changes Lessons User Reviews and Ratings
  - Physical And Chemical Changes Lessons and Bestseller Lists
5. Accessing Physical And Chemical Changes Lessons Free and Paid eBooks
  - Physical And Chemical Changes Lessons Public Domain eBooks
  - Physical And Chemical Changes Lessons eBook Subscription Services
  - Physical And Chemical Changes Lessons Budget-Friendly Options

6. Navigating Physical And Chemical Changes Lessons eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical And Chemical Changes Lessons Compatibility with Devices
  - Physical And Chemical Changes Lessons Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical And Chemical Changes Lessons
  - Highlighting and Note-Taking Physical And Chemical Changes Lessons
  - Interactive Elements Physical And Chemical Changes Lessons
8. Staying Engaged with Physical And Chemical Changes Lessons
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical And Chemical Changes Lessons
9. Balancing eBooks and Physical Books Physical And Chemical Changes Lessons
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical And Chemical Changes Lessons
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical And Chemical Changes Lessons
  - Setting Reading Goals Physical And Chemical Changes Lessons
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical And Chemical Changes Lessons
  - Fact-Checking eBook Content of Physical And Chemical Changes Lessons
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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