## Third Semester B.E. Degree Examination, June/July 2016 Basic Thermodynamics

Time: 3 bes.

Max. Marks:100

Note: 1. Answer FIVE full questions, selecting at least TWO questions from each part.

Use of thermodynamic charts and Tables is permitted.

#### PART - A

- 1 n. Distinguish between:
  - Macroscopie and Microscopie approaches
  - Diathermic and adiabatic walls
  - iii) Intensive and extensive properties
  - iv) Thermal equilibrium and Mechanical equilibrium
  - v) Quasistatic and actual process
  - vi) Point function and path function. (12 Marks)
  - b. The readings t<sub>A</sub> and t<sub>B</sub> of two Celsius thermometers A and B agree at the ice point (0°C) and steam point (100°C), but else where they are related by the equation t<sub>A</sub> = L + mt<sub>B</sub> + mt<sub>B</sub> where t, m and n are constants, when both the thermometers are immersed in a well stirred bath, A registers 51°C, whereas B registers 50°C, Determine the reading on B when A registers 25°C.
    (08 Marks)
- 2 a. Starting from a common stage point, draw the following process on the PV plane and derive an expression for the work done in each case
  - i) Isobaric process
  - ii) Isothermal process
  - ili) Polytropic process
  - Distinguish between heat and work in thermodynamics.

    A spherical balloon has an initial diameter of 25cm and contains air at 1.2 bar. Because of
  - e. A spherical balloon has an initial diameter of 25cm and contains air at 1.2 bar. Because of heating the diameter of the balloon increases to 30cm and during the heating process the pressure is found to be proportional to the diameter. Calculate the work done during the process. (07 Marks)
- State First Law of Thermodynamics for a process and show that energy is a property of the system. (96 Marks)
  - Derive the steady flow energy equation for a single stream of fluid entering and a single stream of fluid leaving the control volume.
    - The following expression gives the specific heat at constant pressure of a gas that undergoes a non-flow constant pressure process  $e_p = \left[ 2.5 + \frac{60}{T + 40} \right] kJ/kg$  K, where T is in °C.
      - The pressure during the process is maintained at 2 bar and the volume changes from 0.3 to 0.56 m³/kg. The corresponding change in temperature is from 25°C to 300°C. Determine
        - i) Work done and heat added
        - ii) Change in internal energy and enthalpy.

(08 Marks)

002 Markst

- State and prove that Kelvin Planck and clausius statements of second Law of Thermodynamics are equivalent. (12 Marks)
  - b. A heat engine working on Carnot cycle absorbs heat from three thermal reservoirs at 1000K, 800K and 600K. The engine does 10kW of network and rejects 400kJ/min of heat to a heat sink at 300K. If the heat supplied by the reservoir at 1000K is 60% of the heat supplied by the reservoir at 600K. Make calculations for the quantity of heat absorbed by each reservoir. (08 Marks)

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A. Ganeshi, G. Balasubramanian

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#### **Table of Contents Vtu Previous Years Question Papers**

- 1. Understanding the eBook Vtu Previous Years Question Papers
  - The Rise of Digital Reading Vtu Previous Years Question Papers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Vtu Previous Years Question Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Vtu Previous Years Question Papers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Vtu Previous Years Question Papers
  - Personalized Recommendations
  - Vtu Previous Years Question Papers User Reviews and Ratings
  - Vtu Previous Years Question Papers and Bestseller Lists
- 5. Accessing Vtu Previous Years Question Papers Free and Paid eBooks

- Vtu Previous Years Question Papers Public Domain eBooks
- Vtu Previous Years Question Papers eBook Subscription Services
- Vtu Previous Years Question Papers Budget-Friendly Options
- 6. Navigating Vtu Previous Years Question Papers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Vtu Previous Years Question Papers Compatibility with Devices
  - Vtu Previous Years Question Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Vtu Previous Years Question Papers
  - Highlighting and Note-Taking Vtu Previous Years Question Papers
  - Interactive Elements Vtu Previous Years Question Papers
- 8. Staying Engaged with Vtu Previous Years Question Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Vtu Previous Years Question Papers
- 9. Balancing eBooks and Physical Books Vtu Previous Years Question Papers
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Vtu Previous Years Question Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Vtu Previous Years Question Papers
  - Setting Reading Goals Vtu Previous Years Question Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Vtu Previous Years Question Papers
  - Fact-Checking eBook Content of Vtu Previous Years Question Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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