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Fission and Fusion

11. The number of neutrons the atom's nucleus changes in a fission process. The number _____ decreases _____ change.
12. How is the mass conserved? Mass defect is the difference between the mass of the separate particles before and the products in a nuclear reaction.
13. Calculate the energy released when 1 kg of ^{235}U is fissioned at 200 MeV per fission. Assume the mass of ^{235}U is 235 g/mol. Fission is the splitting of one nucleus and is used for nuclear power plants.
14. Calculate the energy released when 1 kg of ^{235}U is fissioned at 200 MeV per fission. Assume the mass of ^{235}U is 235 g/mol. Fission is the splitting of one nucleus and is used for nuclear power plants.

$$\frac{235 \text{ g}}{1 \text{ mol}} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.000425 \text{ kg}$$

$$E = mc^2 = (0.000425 \text{ kg})(3.00 \times 10^8 \text{ m/s})^2 = 3.83 \times 10^{13} \text{ J}$$

15. Balance and find the following nuclear reactions. Calculate how many neutrons are created.

_____ neutrons _____ $^{235}\text{U} \rightarrow ^{141}\text{Ba} + ^{92}\text{Kr} + \underline{3}\text{n}$

_____ neutrons _____ $^{235}\text{U} + ^1_0\text{n} \rightarrow ^{141}\text{Ba} + ^{92}\text{Kr}$

_____ neutrons _____ $^{235}\text{U} \rightarrow ^{141}\text{Ba} + ^{92}\text{Kr} + \underline{3}\text{n}$

_____ neutrons _____ $^{235}\text{U} \rightarrow ^{141}\text{Ba} + ^{92}\text{Kr} + \underline{3}\text{n}$

_____ neutrons _____ $^{235}\text{U} + ^1_0\text{n} \rightarrow ^{141}\text{Ba} + ^{92}\text{Kr} + \underline{3}\text{n}$ (How many neutrons are created?)

16. a. Calculate the following is an example of a nuclear reaction?

(a) The nuclear mass

(c) nuclear fission/fusion

(b) Nuclear power plants

(d) Radioactive decay

17. a. The mass of the reaction and mass in a power plant reaction controlled by the bar

(a) mass of the reaction

(c) change in concentration of neutrons

(b) heat output and efficiency around the fuel rods

(d) different fuel reactions

18. Describe how applications of nuclear technology is helping us live better.

Radioactive Dating - The age of an object is determined based on the presence of specific radioactive isotopes.

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Zhe Xu



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