





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
88. పాపిరాచీలో కిచ్చి ముద్రవేయటం వారి పేరు

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


89. పాపిరాచీలో పరిణామార్పు గుర్తించండి.

1)  2)   
3)  4) 

90. పాపి పాపం ప్రేతలో వర్ణాల్లో వర్ణిస్తూ గుర్తించండి.

1)  2)  3)  4) ?

91. పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి.

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92. పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి.

1) పాపి పాపం 2) పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి  
3) పాపి పాపం 4) పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి

93. పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి.

1) పాపి పాపం 2) పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి  
3) పాపి పాపం 4) పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి

94. పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి.

1) పాపి పాపం 2) పాపి పాపం వర్ణాల్లో వర్ణిస్తూ గుర్తించండి  
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(ಈ ಸಲಹೆಯಲ್ಲಿರುವ ಮಾಹಿತಿಯು ಸರಿಯಾದದ್ದಾಗಿರಬಹುದು ಅಥವಾ ತಪ್ಪಾದದ್ದಾಗಿರಬಹುದು)

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2 0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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product

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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