WK.	1.55	TOPICS-TOPIC	OBJECTIVES	L'ACTIVITIES	L/T AIDS	REFERENCE	REMARKS
1	100		SCHOOL OPENING		7		
	2	Structure of the atom and the periodic table - Structure of the atom	By the end of the lesson, the learner should be able to: - Define atom - Name the subatomic particles and Where found - Describe the structure of the atom and label it	Notes taking Discussion Drawing and labeling the atom	Chart of the structure of the atom	KLB BK 2 Pg I	
	3	Atomic number and Atomic mass	By the end of the lesson, the learner should be able to: - Define atomic number and atomic mass - Do simple calculations on how to find either of the given one of them	- Discussion - Notes taking - Calculation	Past questions	KLB BK 2 Pg 2-3	
	*	Isotopes	By the end of the lesson, the learner should be able to: - Define isotopes - Name some elements that exhibit lisotopes - Name isotopes of hydrogen	Discussion Notes taking Calculation	- Chalkboard - Charts	KLB BK 2 Pg 4	
2	182	Energy levels and Electron arrangement	By the end of the lesson, the learner should be able to:- - Write the electronic arrangement of The 1st 20 elements - State the importance of the number of energy levels as well as the number of electrons in the outermost energy level	- Discussion - Notes taking - Writing of the E.C	Charts of E.C of the 1" 20 elements	KLB BK 2 Pg 4-7	
	4	Development of the Periodic table	By the end of the lesson, the learner should be able to: - Build up the periodic table for the 1st 20 elements based on the no. of energy levels and no. of electrons in the outermost energy level. - Explain the position of an element in The table	Discussion Notes taking Class activity	- Mamila - Plasticine - Ruler - Felt pens - Charts of the 1st 20 elements - Charts of the 1st 20 elements	KLB BK 2 Pg 8-9	
3	18:2	Relative atomic mass	By the end of the lesson, the learner				

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D Keegan

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