











AQA Physics 1 Scheme of Learning 2011- onwards

Objectives	Starter	Establishing	Development	Plenary	Practicals	Syllabus content
1 Kinetic Theory All Describe the properties of the 3 states of matter Most Represent the 3 states of matter with a particle diagram, explain how energy affects the particles & explain how energy changes can cause a change of state Some Compare and contrast the 3 states of matter in terms of particles, vibrations and space Useful websites http://www.aqa.org.uk/Physics/Theory	Recap the properties of the 3 states of matter <ul style="list-style-type: none"> Group discussion Brain whiteboards Brain storm or concept map  whole class feedback/collecting ideas  provide key words for students to classify into the 3 states of matter  focus on particle diagrams	Discuss the amount of energy in each state of matter <ul style="list-style-type: none"> Card sort of the main properties Fill in table with similarities and differences Show animations for each state  pupils share ideas with each other before whole class feedback  provide diagrams for them to annotate  explain why energy causes the particle diagram to look different	Discuss the processes involved in evaporation and cooling and how this can be represented through particle diagrams. Discuss how the properties change <ul style="list-style-type: none"> Practical on changing state Drawing particle diagrams showing the what happens during a change of state Identifying how the properties change during a change of state Compare and contrast the properties of each change of state Practical  teacher to circulate checking progress and Questioning  provide practical method and table for experiments. Ask pupils to focus on the 3 states rather than the changes of state  plan a fair experiment to investigate the effect of heating and cooling. Plan, assess and improve method before starting the experiment. Evaluate the experiment. Compare and contrast the 3 states	<ul style="list-style-type: none"> True or false statements about each state and change of state processes  go through answers on the board  syllabus statement checklist  provide clues for the difficult statements (pictures etc.)  write their own true and false statements on mini whiteboards	<ul style="list-style-type: none"> Order P1 Lesson 1a demonstration using balls in Perspex boxes. Order P1 Lesson 1b kinetic theory demo (balls in trays) 	<ul style="list-style-type: none"> There are 3 states of matter: Solid, Liquid & Gas The atoms in solids, liquids & gases have different amounts of energy In solids the particles are packed very close together. They vibrate about fixed positions In liquids the particles are close together but not as close as they are in solids. They can move around in any direction and are not fixed in position. In gases the particles are very far apart with large distances between them. They move around very quickly in all directions You can change the state of a substance by heating or cooling it

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