



# CHRIST THE REDEEMER SECONDARY SCHOOL

3, Morenikeji Street, Behind Jolad Hospital New Garage, Gbagada, Lagos

**2024/2025 Second Term Examination (MOCK)**

**SUBJECT: CHEMISTRY**

**CLASS:**

**SSS THREE**

## SECTION A - OBJECTIVE TEST

- The constituent common to duralumin and alnico is
  - Co
  - Mn
  - Al
  - Mg
- The monomer of nylon is?
  - hydroxyl benzene
  - hexamethanoic acid
  - benzene-1, 4-dicarboxylic acid
  - ethane-1, 2-acid
- A compound that gives a brick-red colour to a non-luminous flame is likely to contain
  - Aluminium ions
  - Copper ions
  - Sodium ions
  - Calcium ions
- In the electrolytic extraction of calcium ions from Calcium Chloride, the cathode is
  - Iron
  - Zinc
  - Graphite
  - Platinum
- A few drops of NaOH solution was added to an unknown salt forming a white precipitate which is insoluble in excess solution, the cation likely present is
  - $\text{Al}^{3+}$
  - $\text{Zn}^{2+}$
  - $\text{Pb}^{2+}$
  - $\text{Ca}^{2+}$
- Ethyne is passed through a hot tube containing organo-nickel catalyst to produce
  - benzene
  - isoprene
  - polythene
  - ethanol
- Due to the unstable nature of ethyne, it is stored by dissolving in
  - propanone
  - ethane-1, 2-diol
  - propanol
  - ethanoic acid
- The polymer used in making car rear lights is
  - polyacrylonitrile
  - perspex
  - bakelite
  - polystyrene
- $\text{CH}_3\text{COO C}_2\text{H}_5 + \text{H}_2\text{O} \xrightarrow{\text{H}^+} \text{CH}_3\text{COOH} + \text{CH}_3\text{CH}_2\text{OH}$ . The purpose of  $\text{H}^+$  in the reaction above is to
  - decrease the rate of the reverse reaction
  - increase the yield of the product
  - maintain the solution at a constant pH
  - increase the rate of the hydrolysis
- A hydrocarbon has an empirical formula  $\text{CH}$  and a vapour density of 39. Determine its molecular formula
  - $\text{C}_6\text{H}_6$
  - $\text{C}_2\text{H}_2$
  - $\text{C}_3\text{H}_4$
  - $\text{C}_3\text{H}_4$
- The process of converting linear alkanes to branched chain and cyclic hydrocarbon by heating in the presence of a catalyst to improve the quality of petrol is referred to as
  - blending
  - refining
  - cracking
  - reforming
- The rate of diffusion of a gas Y is twice that of Z. If the relative molecular mass of Y is 64 and the two gases diffuse under the same condition, find the relative molecular mass of Z
  - 8
  - 16
  - 32
  - 4
- The radioisotope used in industrial radiography for the rapid checking of faults in welds and casting is
  - Cobalt-60
  - Iodine-131

# Ss2 Mock Question On Chemistry 2015

**Yicheng Fang**



## **Ss2 Mock Question On Chemistry 2015:**

The book delves into Ss2 Mock Question On Chemistry 2015. Ss2 Mock Question On Chemistry 2015 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Ss2 Mock Question On Chemistry 2015, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Ss2 Mock Question On Chemistry 2015
- Chapter 2: Essential Elements of Ss2 Mock Question On Chemistry 2015
- Chapter 3: Ss2 Mock Question On Chemistry 2015 in Everyday Life
- Chapter 4: Ss2 Mock Question On Chemistry 2015 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Ss2 Mock Question On Chemistry 2015. The first chapter will explore what Ss2 Mock Question On Chemistry 2015 is, why Ss2 Mock Question On Chemistry 2015 is vital, and how to effectively learn about Ss2 Mock Question On Chemistry 2015.

3. In chapter 2, the author will delve into the foundational concepts of Ss2 Mock Question On Chemistry 2015. The second chapter will elucidate the essential principles that must be understood to grasp Ss2 Mock Question On Chemistry 2015 in its entirety.

4. In chapter 3, the author will examine the practical applications of Ss2 Mock Question On Chemistry 2015 in daily life. The third chapter will showcase real-world examples of how Ss2 Mock Question On Chemistry 2015 can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Ss2 Mock Question On Chemistry 2015 in specific contexts. This chapter will explore how Ss2 Mock Question On Chemistry 2015 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Ss2 Mock Question On Chemistry 2015. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Ss2 Mock Question On Chemistry 2015.

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