

STAAR CHEMISTRY REFERENCE MATERIALS



OTHER FORMULAS

$$\text{Density} = \frac{\text{mass}}{\text{volume}}$$

$$D = \frac{m}{V}$$

$$\text{Percent error} = \left(\frac{\text{accepted value} - \text{experimental value}}{\text{accepted value}} \right) (100)$$

$$\text{Percent yield} = \left(\frac{\text{actual yield}}{\text{theoretical yield}} \right) (100)$$

CONSTANTS AND CONVERSIONS

$$\text{Avogadro's number} = 6.02 \times 10^{23} \text{ particles per mole}$$

$$h = \text{Planck's constant} = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$$

$$c = \text{speed of light} = 3.00 \times 10^8 \frac{\text{m}}{\text{s}}$$

$$K_w = \text{ionization constant of water} = 1.00 \times 10^{-14} \left(\frac{\text{mol}}{\text{L}} \right)^2$$

$$\alpha \text{ particle } (\alpha) = {}^4_2\text{He} \quad \beta \text{ particle } (\beta) = {}^0_{-1}\text{e} \quad \text{neutron} = {}^1_0\text{n}$$

$$\text{standard temperature and pressure (STP)} = 0^\circ\text{C and } 1 \text{ atm}$$

$$0^\circ\text{C} = 273 \text{ K}$$

$$\text{volume of ideal gas at STP} = 22.4 \frac{\text{L}}{\text{mol}}$$

$$1 \text{ cm}^3 = 1 \text{ mL} = 1 \text{ cc}$$

$$1 \text{ atm} = 760 \text{ mm Hg} = 101.3 \text{ kPa}$$

$$R = \text{ideal gas constant} = 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{mol} \cdot \text{K}} = 8.31 \frac{\text{L} \cdot \text{kPa}}{\text{mol} \cdot \text{K}} = 62.4 \frac{\text{L} \cdot \text{mm Hg}}{\text{mol} \cdot \text{K}}$$

$$1 \text{ calorie (cal)} = 4.18 \text{ joules (J)}$$

$$1000 \text{ calories (cal)} = 1 \text{ Calorie (Cal)} = 1 \text{ kilocalorie (kcal)}$$

RULES FOR SIGNIFICANT FIGURES

1. Non-zero digits and zeros between non-zero digits are always significant.
2. Leading zeros are not significant.
3. Zeros to the right of all non-zero digits are only significant if a decimal point is shown.
4. For values written in scientific notation, the digits in the coefficient are significant.
5. In a common logarithm, there are as many digits after the decimal point as there are significant figures in the original number.

Staar Chemistry Practice Test B Answers

Karin Nielsen-Saines



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web glass in buildings as1288 2006 residential 1 version 1 released june 2019 the agwa has prepared this residential building summary on key areas of as1288 2006 glass in buildings selection and installation we urge you to be familiar with the standard in its entirety this fact sheet is intended to

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web jan 16 2006 this standard sets out procedures for the selection and installation of glass in buildings subject to wind loading human impact and special applications such as overhead glazing balustrades and glass assemblies glass strength requirements are given for glazing based on the tensile stresses developed on the surface of the

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web description you may need to comply with this standard if you intend to operate a caravan park or camping ground this standard sets out procedures for the selection and installation of glass in buildings subject to wind loading human impact and special applications such as overhead glazing balustrades and glass assemblies

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web the revised as1288 2021 glass in building standard released on 26 6 21 has further clarified these stringent requirements including renaming the balustrade section to now be known as barrier to fall in line with national construction code ncc descriptions