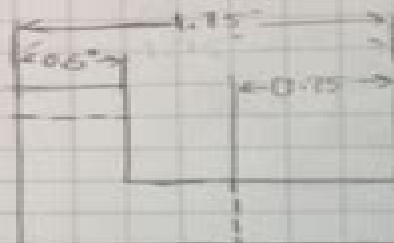
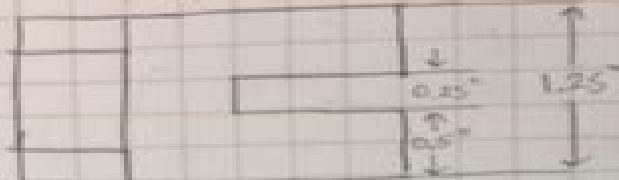


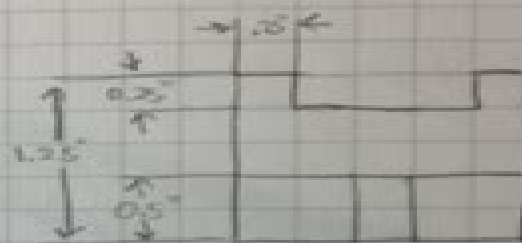
Activity 5.6 Physical Property Analysis

0.25" Spacing

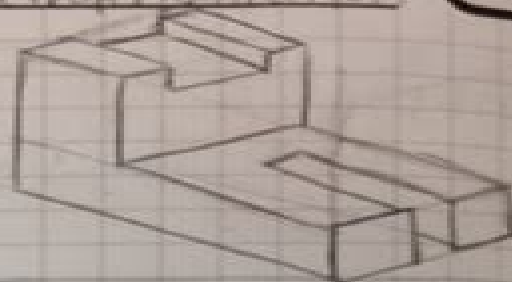
TOP



front



side



$$1. [(.5 \cdot 1.25 \cdot 1.25) - (.25 \cdot .5 \cdot .75)] + [(.5 \cdot 1.25 \cdot 1.25) - (.25 \cdot .75 \cdot .5)]$$

$$= 1.375 \text{ in}^3$$

$$2. 2.7 \text{ g/cm}^3 \text{ (education.jlab.org)}$$

$$3. \frac{2.7}{1} \cdot \frac{0.0220462}{0.0610257} = \frac{48}{1000}$$

$$= 0.098 \text{ lb/in}^3$$

$$4. 0.098 \cdot \frac{1}{135}$$

$$= 0.135 \text{ lb/in}^3$$

$$5. [(.75 \cdot 1.25) - (.25 \cdot .75)] \cdot 2$$

$$+ 1.25 \cdot 1.25 = 1.5625 = 1.375$$

$$+ 1.25 \cdot 1.25$$

$$+ 1.375 \cdot 2 = 10.25 \text{ in}^3$$

$$6. \text{Density} = 2.7 \text{ g/cm}^3$$

$$\text{Mass} = 0.154 \text{ lb}$$

$$\text{SA} = 10.25 \text{ in}^2$$

$$\text{Volume} = 1.375 \text{ in}^3$$

$$7. (2000 \cdot 10.25) / 14400$$

$$= 22.135 \text{ g/in}^3$$

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W. Verhaegh

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