

CLASS - X

LAKHMIR SINGH &
MANJEET KAUR

S. CHAND

VERY SHORT Q. NO 1-16
SOLUTION

Very Short Answer Type Questions

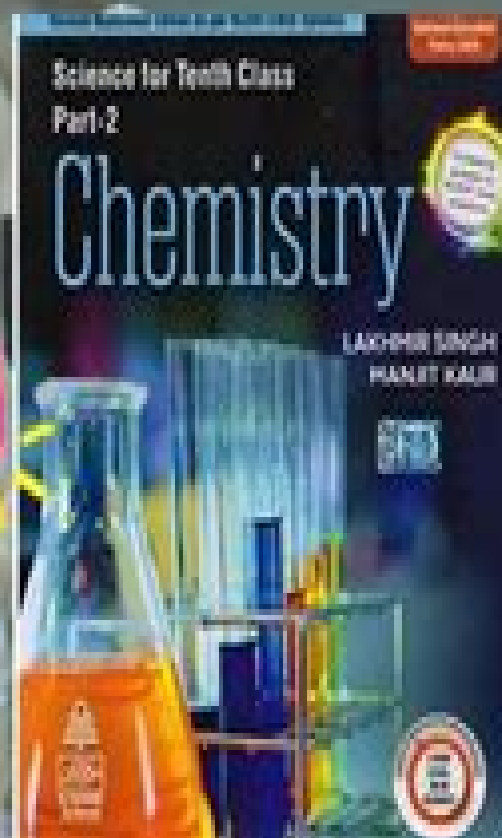
1. Why is respiration considered an exothermic process?
2. On what basis is a chemical equation balanced?
3. What happens chemically when quicklime is added to water filled in a bucket?
4. Why should magnesium ribbon be cleaned before burning in air?
5. State whether the following statement is true or false:
A chemical equation can be balanced only by altering the formulae of a reactant or product.
6. What is wrong with the following chemical equation?



Correct and balance it.

7. What does the symbol (aq) represent in a chemical equation?
 8. Why is photosynthesis considered an endothermic reaction?
 9. How will you indicate the following effects in a chemical equation?
- (a) A solution made in water
 - (b) Exothermic reaction
 - (c) Endothermic reaction
10. Translate the following statements into chemical equations and then balance the equations:
 - (a) Hydrogen sulphide gas burns in air to give water and sulphur dioxide.
 - (b) Phosphorus burns in oxygen to give phosphorus pentoxide.
 - (c) Carbon disulphide burns in air to give carbon dioxide and sulphur dioxide.

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