

Prentice Hall Physical Science

Chapter 4 Atomic Structure

4.1 Studying Atoms

- A. Ancient Greek Models of Atoms
 - Democritus believed all matter was made of extremely small particles that could not be divided
 - he called the particles atoms (from the Greek atomos meaning indivisible)
 - he thought there were different types of atoms with different properties (liquids had round, smooth atoms and solids were rough and prickly)
- B. Dalton's Atomic Theory
 - he proved atoms exist by discovering that compounds have a fixed composition
 - Dalton's Theory:
 - all elements are composed of atoms
 - all atoms of the same element have the same mass, and atoms of different elements have different masses
 - compounds contain atoms of more than one element
 - in a particular compound, atoms of different element always combine in the same way
 - he pictured atoms as solid spheres
 - in time it was found that not all of this was completely correct
- C. Thomson's Model of the Atom
 - some objects have a positive or negative charge and objects with the same charge repel and objects with opposite charges attract
 - Thomson used an electric current to learn more about atoms
 - he passed an electric current through a gas and a glowing beam appeared
 - the beam was repelled by a negatively charged metal plate (like charges repel and opposite charges attract)
 - Thomson concluded that the beam must be made of negatively charged particles, smaller than atoms.
 - since the atom was known to be electrically neutral, he developed a model in which negative charges were evenly scattered throughout an atom filled with a positively charged pudding like material
 - called the "Plum Pudding" model (or chocolate chip cookie dough)
- D. Rutherford's Atomic Theory
 - Rutherford discovered positively charged, fast moving particles called alpha particles
 - he shot some of these particles a piece of gold foil, thinking they would go straight through
 - when he actually performed the experiment, he found that the positively charged alpha particles were greatly deflected by the gold (figure 7, p. 104)
 - he determined that the alpha particles must have come very close to another positively charged particle causing them to be repelled and put off course
 - this led him to believe that the positive charge in an atom is not evenly distributed, but concentrated in a very small central area he called the nucleus
 - his model of the atom is a small, positively charged nucleus surrounded by negatively charged particles

Prentice Hall Physical Science Chapter 11 Review Answers

Léo Pomar



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Teaching Secondary Science Karen Mesmer, Enya Granados, Kevin Gant, Laura Shafer, Ayanna D. Perry, 2024-01-16 Your guide to grow and flourish as a science teacher The past two decades have seen a paradigm shift in science education thanks in large part to the Next Generation Science Standards NGSS which advocate a move away from procedural lab investigations and direct instruction and toward increased emphasis on reasoning sensemaking phenomena exploration problem solving and collaboration Under this new paradigm students are learning real science as scientists practice it so that more and more students are actively investigating questions and pursuing solutions of their own making As part of the Five to Thrive series for early career educators this comprehensive guide provides those who are new to teaching science as well as seasoned teachers looking to enhance their practice the fundamentals to develop best teaching practices that reflect their students experiences and requirements Written by experienced science educators Answers to Your Biggest Questions About Teaching Secondary Science provides practical guidance on successful strategies and techniques for teaching science in a way that gives every student the opportunity to learn grow and achieve at high levels while providing opportunities to develop their agency and authority in the classroom ultimately resulting in a positive science identity The book is organized around five overarching questions and answers that will help you most thrive in your secondary science classroom How do I build a positive science community How do I structure organize and manage my science class How do I engage my students in science How do I help my students talk about science How do I know what my students know and how can I use that information to plan and move them forward The book concludes with a sixth question Where do I go from here that provides guidance for growing your practice over time including discussions on self care advocating for students and an extensive discussion on growing your professional network Woven throughout you ll find helpful sidebar notes on fostering identity and agency access and equity teaching in different settings and invaluable resources for deeper learning Strive to become the best science educator you can be your students are counting on it **Saturday Review** ,1862 *The Saturday Review of Politics, Literature, Science, Art, and Finance* ,1862 Resources in Education ,1991-03 **Research in Education** ,1974

Science Insights DiSpezio Diaz,1996 **Athenaeum** ,1845 **About Criminals** Mark Pogrebin,2011-12-09 This book presents students with recent and important research on criminal behavior The articles in this anthology all based on actual field studies provide the reader with a realistic portrayal of what actual offenders say about crime and their participation in it The offenders voices along with the researchers analyses offer students a real life view of what how and why various criminals behave the way they do **Athenaeum and Literary Chronicle** ,1838 The Athenaeum ,1859 **Biomedical Science and Clinical Foundations** Cynthia A. Standley,2025-09-26 This book teaches medical students basic science and clinical concepts utilizing over 30 cases that cover all major organ systems Case based learning has become a core way to teach the application of foundational material in the basic and clinical sciences in the pre clerkship years Medical schools curriculums tend to be traditional organ systems based or problem based There are many discipline based textbooks for

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