

Class Task Revision Chemistry paper 1

1.

The table shows information about elements.

atomic symbol	number of electrons in	
	one atom of the element	one ion of the element
Li	3	2
Mg	12	10
Al	13	10
Cl	17	18
K	19	18
Ca	20	18

(a) (i) Which atom loses three electrons to form an ion?

Circle the correct answer.

1. Li Mg Al Cl K Ca [1]

(ii) Which atom forms a negative ion?

.....

Explain your answer.

.....

.....

[2]

(iii) Two elements in the table are in Group 1.

Write down the atomic symbols of these **two** elements.

Use the Periodic Table on page 18 to help you.

..... and

[1]

Previous Chemistry 2quiz

CL Gary



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