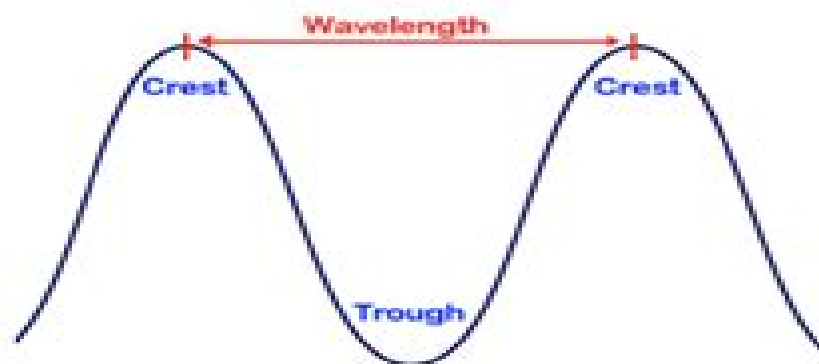


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Wavelength (λ) - Units: m, cm, nm

Frequency (ν) - the # of waves that pass a given point in one second.

Units: Hz, sec^{-1} , waves/sec, 1/sec

The relationship between frequency (f) and wavelength (λ) of a wave is described by the equation

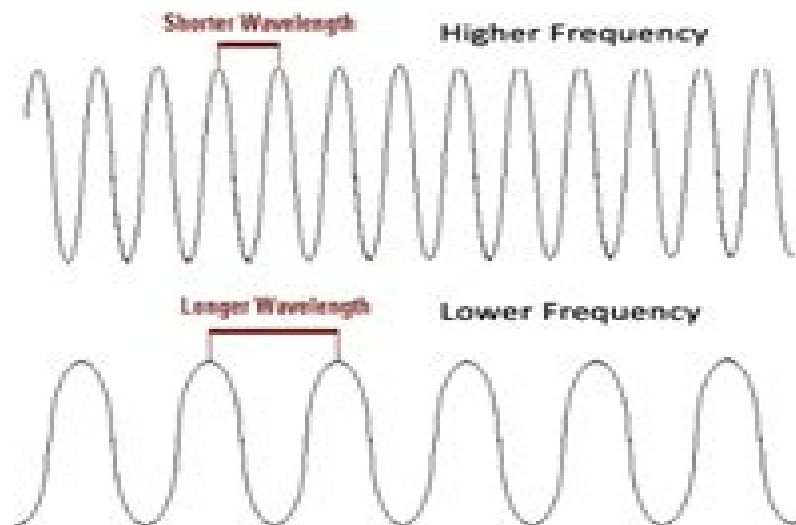
$$\text{Waves speed} = \text{frequency} \times \text{wavelength} \quad \text{OR} \quad c = \nu \lambda$$

All electromagnetic radiation propagates at a speed of $c = 3.00 \times 10^8$ m/sec (approximately 186,000 miles per second) this is called the speed of light!

When solving problems using c (the speed of light) ν MUST be in Hz or sec^{-1} and λ MUST be in meters.

Problems

1. What is the wavelength of radiation with a frequency of $6.24 \times 10^{13} \text{ sec}^{-1}$?
2. What is the frequency of radiation with a wavelength of 2.20×10^{-6} nm? (1 nm = 10^{-9} m or $10^9\text{nm}=1\text{m}$)
3. A television relay transmitter operates at a frequency of 927.9 MHz. Calculate the wavelength of the signal in nanometers. Is this visible light? (1 MHz = 1,000,000 Hz)



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Baby Professor



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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more *Invisible Universe* Stephen M. Pompea, Alan Gould, Lincoln Bergman, 2002 The 5 class

sessions of 45 60 minutes each deepen student understanding of the electromagnetic spectrum enabling students to detect and consider wavelengths other than visible light Activities feature energy stations including infrared TV remote microwave pager ultraviolet black light and other devices Students come up with their own tests to see what blocks each wavelength and what does not They learn how these other wavelengths can be used to see things we cannot see with our eyes **The**

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Electromagnetic Spectrum Ginger Butcher, 2016 *Tour of the Electromagnetic Spectrum* Ginger Butcher, Claire L. Parkinson, Edward J. Wollack, 1915-09-30 This publication introduces the electromagnetic energy and the properties of electromagnetic waves across the spectrum Each wavelength is shown with stunning NASA imagery with explanations of how NASA instruments collect data to help study Earth and space The Electromagnetic Spectrum (Set) Alison

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Lesson Plan on the Electromagnetic Spectrum , The Johns Hopkins University Applied Physics Laboratory presents a lesson plan entitled Lesson Plan on the Electromagnetic Spectrum This lesson is suitable for use with students in grades 8-12 The students will explore the various wavelengths in the electromagnetic spectrum The laboratory offers this lesson plan as part of the Educator's Guide to Near Earth Asteroid Rendezvous NEAR

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