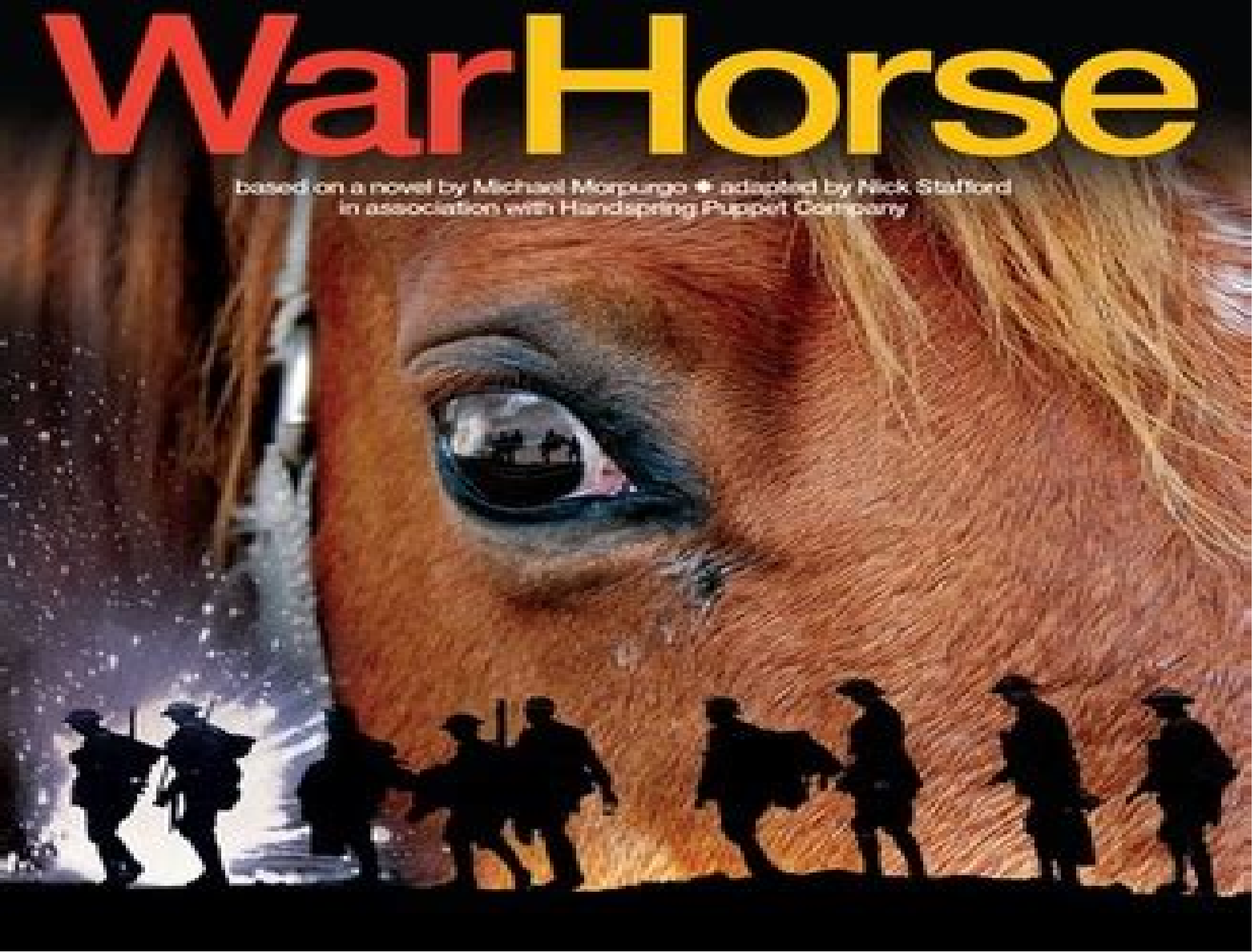


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# War Horse Film Study Guide

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web jul 12 2019 when we neutralize a weak acid with a strong base we get a salt that contains the conjugate base of the weak acid this conjugate base is usually a weak base for example sodium acetate  $\text{NaCH}_3\text{CO}_2$  is a salt formed by the reaction of the weak acid acetic acid with the strong base sodium hydroxide

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web summary an arrhenius acid produces  $\text{H}^+$  and an arrhenius base produces  $\text{OH}^-$  in aqueous solutions acids taste sour may sting and neutralize bases bases taste bitter feel slippery and neutralize acids naming acids binary acids contain a single anion  $\text{HX}$  to name hydro anion with  $\text{H}^+$  end acid

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web definition of acids and bases identifying them along with the conjugate acid and conjugate base ph of strong and weak acids and bases the ph and poh relationship  $K_a$  and  $K_b$  p  $K_a$  and  $K_a$  correlation the acid base properties of salts identifying the acidity of the salt based on the acid and the base it is prepared from and calculating the