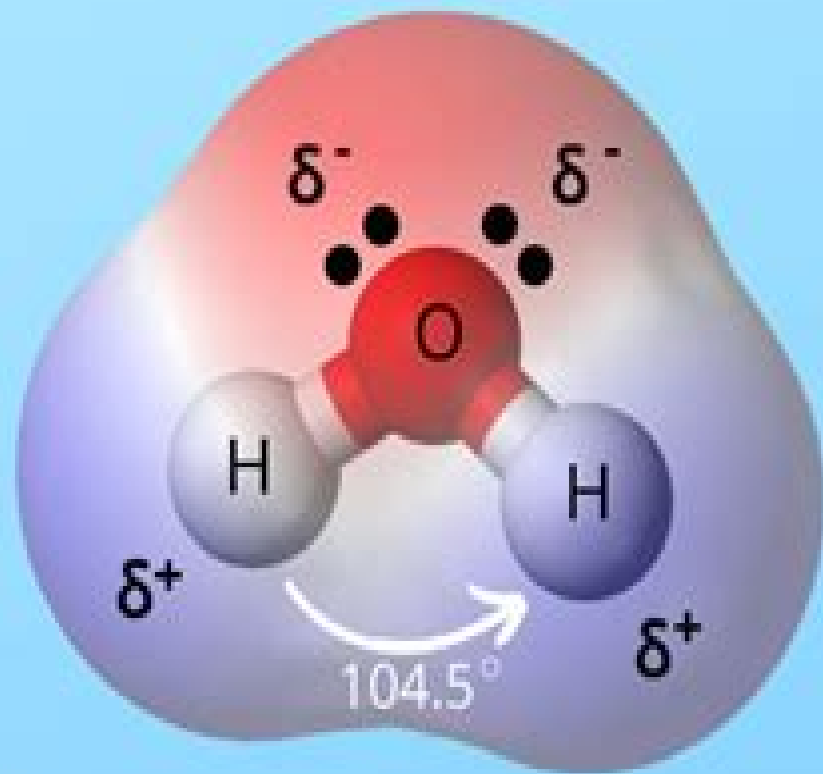


Why Water Is a Polar Molecule

- Water is polar because oxygen and hydrogen have different electronegativity values.
- Oxygen has two lone electron pairs that repel each other and the electrons bonded to the hydrogen atoms.
- This gives the water molecule a bent shape.
- The oxygen side has a partial negative charge; the hydrogen side has a partial positive charge.



Water Is A Polar Liquid And An Excellent

O García



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