

17

WATER AND AQUEOUS SYSTEMS

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SECTION 17.1 LIQUID WATER AND ITS PROPERTIES (pages 475–478)

This section explains the hydrogen bonding that occurs in water molecules and how high surface tension and low vapor pressure can be explained in terms of hydrogen bonding.

► The Water Molecule (pages 475–476)

1. What unique molecular compound is the foundation of all life on Earth?

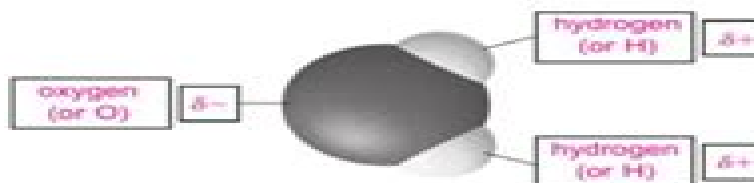
water

2. Approximately what fraction of Earth's surface is covered in water? 3/4

3. Circle the letter next to each sentence that is true concerning water molecules.

- a. Each O — H covalent bond in a water molecule is nonpolar.
☒ b. In a water molecule, the less electronegative hydrogen atoms acquire a slight positive charge and the oxygen atom acquires a slight negative charge.
☒ c. Because the water molecule has an H — O — H bond angle of 105° , the molecule as a whole is polar.

4. The diagram below depicts a water molecule. Complete the labels showing the locations of the hydrogen atoms, the oxygen atom, and the regions of positive and negative charge.



5. List five important properties of water that result from hydrogen bonding.

- a. high surface tension
 b. low vapor pressure
 c. high specific heat capacity
 d. high heat of vaporization
 e. high boiling point

Water And Aqueous Systems Answers

Lingsheng Yao



Water And Aqueous Systems Answers:

Water And Aqueous Systems Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has are more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Water And Aqueous Systems Answers**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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