

First Year Higher Secondary Examination

PART – III

CHEMISTRY

Maximum: 60 Scores

Time: 2 Hours

Cool-off time: 15 Minutes

1. (a) Determine the number of moles present in 0.55 mg of electrons.
 (i) 1 mole (ii) 2 mole (iii) 1.5 mole (iv) 0.5 mole (1)
 (b) Give the empirical formula of the following. (2)
 $C_8H_{12}O_6$, C_6H_6 , CH_3COOH , $C_6H_6Cl_6$
- (c) Two elements, carbon and hydrogen combine to form C_2H_6 , C_2H_4 and C_2H_2 , Identify the law illustrated here. (1)
2. (a) (i) Write the electronic configuration of chromium ($Z = 24$). (1)
 (ii) Find the number of electrons in the subshells with azimuthal quantum number $l = 2$. (1)
 (iii) Represent the orbital with quantum numbers $n = 1$ and $l = 0$ (1)
 (b) Give the mathematical representation of Heisenberg's uncertainty principle and its one important significance. (2)
3. Electron gain enthalpy is one of the important periodic property.
 (a) Define electron gain enthalpy. (1)
 (b) Explain any two factors affecting electron gain enthalpy. (2)
 (c) Write the oxidation state and co-valency of Al in $(AlF_6)^{3-}$. (1)
4. The geometry of the molecule is decided by type of hybridization.
 (a) Discuss the shape of PCl_5 molecules using hybridization. (2)
 (b) Give the reason for the high reactivity of PCl_5 . (2)
 (c) Isoelectric species have the same bond order. Among the following, choose the pair having same bond order.
 CN^- , \bar{O}_2 , NO^+ , CN^+ . (1)

Polytechnic 1st Year Chemistry Question Paper

Clifford Lane



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1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Polytechnic 1st Year Chemistry Question Paper
- Chapter 2: Essential Elements of Polytechnic 1st Year Chemistry Question Paper
- Chapter 3: Polytechnic 1st Year Chemistry Question Paper in Everyday Life
- Chapter 4: Polytechnic 1st Year Chemistry Question Paper in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Polytechnic 1st Year Chemistry Question Paper. This chapter will explore what Polytechnic 1st Year Chemistry Question Paper is, why Polytechnic 1st Year Chemistry Question Paper is vital, and how to effectively learn about Polytechnic 1st Year Chemistry Question Paper.
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5. In chapter 4, the author will scrutinize the relevance of Polytechnic 1st Year Chemistry Question Paper in specific contexts. The fourth chapter will explore how Polytechnic 1st Year Chemistry Question Paper is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Polytechnic 1st Year Chemistry Question Paper. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Polytechnic 1st Year Chemistry Question Paper.

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