



Neutralization Reactions Worksheet

- 1. Write the balanced chemical equations for the neutralization reactions between:
 - a) HI and NaOH

$$HI + NaOH \rightarrow H_2O + NaI$$

b) H₂CO₃ and Sr(OH)₂

$$H_2CO_3 + Sr(OH)_2 \rightarrow 2 H_2O + SrCO_3$$

c) Ca(OH)₂ and H₂PO₄

$$3 \text{ Ca}(OH)_2 + 2 \text{ H}_3PO_4 \rightarrow 6 \text{ H}_2O + \text{Ca}_3(PO_4)_2$$

d) hydrobromic acid and barium hydroxide

e) zine hydroxide and nitrie acid

$$Zm(OH)_2 + 2 HNO_3 \rightarrow 2 H_2O + Zm(NO_3)_2$$

aluminum hydroxide and hydrochloric acid

$$AI(OH)_3 + 3 HCI \rightarrow 3 H_2O + AICI_3$$

2. Complete and balance the following equations representing neutralization reactions:

Robert A. Richardson

Proton Chemistry Robert A. Richardson,1976 Alberta Authorized Resource for grade 12 ca 1980 1994

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