

**C11- PHYSICAL SCIENCE**  
 Third Monthly Test Reviewer

Coverage: **MODULE 1**  
 UNIT 1: Atomic Theory  
 UNIT 2: The Discovery and Synthesis of Chemical Elements

**UNIT 1: Atomic Theory**

- Atoms are composed of **protons, neutrons, and electrons.**



Table 2.3 Properties of Subatomic Particles

Particle	Symbol	Mass (kg)	Relative Mass (proton =1)	Relative Charge
Proton	$p^+$	$1.673 \times 10^{-27}$	1	+1
Neutron	$n^0$	$1.675 \times 10^{-27}$	1	0
Electron	$e^-$	$9.109 \times 10^{-31}$	0.00055	-1

- An atom is mostly empty space with a highly dense, **positive** center called **nucleus**.
- For a neutral atom, the **number of protons is equivalent to the number of electrons.**  
*(Lithium has 3 protons therefore it also has 3 electrons)*
- The **atomic number** of an element is also equivalent to its number of **protons.**  
*(Nitrogen has an atomic number of 7 therefore it has 7 protons in its nucleus)*
- The **atomic mass** (mass number) is primarily determined by the number of protons and **neutrons.**

- Isotopes** are atoms with the **same atomic number** but different mass numbers. They **differ in the number of neutrons.**  
*(Carbon-12, Carbon-13, Carbon-14)*
- Ion** is an element or atom that has a **positive or negative net charge.** It is charged because the number of electrons does not equal the number of protons in the atom.

Two types:

- Cation**- positive ion (loses electron)
- Anion**- negative ion (gains electron)



- Take note that when an **atom is not electrically neutral**, it means that it does not have the same number of protons and electrons.
- When an atom or element is **positively charged**, it means that it has **more protons** in its nucleus.

**Historical Development of Atom**

- Democritus & Leucippus:**  
**Atomism:** All matter is made up of very small, indivisible particles called atoms (atomos).
- Aristotle:** Four elements in nature (air, fire, water, and earth)
- John Dalton: Billiard Ball Model/Solid Sphere Model:**  
**Atomic Theory:** All matter is made up of very small, indivisible particles called atoms.

# Physical Science Gradeexemplar Caps 2014

**Brian Hopkins,Elena Geangu,Sally  
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## **Table of Contents Physical Science Gradeexemplar Caps 2014**

1. Understanding the eBook Physical Science Gradeexemplar Caps 2014
  - The Rise of Digital Reading Physical Science Gradeexemplar Caps 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Gradeexemplar Caps 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Gradeexemplar Caps 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Gradeexemplar Caps 2014
  - Personalized Recommendations
  - Physical Science Gradeexemplar Caps 2014 User Reviews and Ratings
  - Physical Science Gradeexemplar Caps 2014 and Bestseller Lists
5. Accessing Physical Science Gradeexemplar Caps 2014 Free and Paid eBooks
  - Physical Science Gradeexemplar Caps 2014 Public Domain eBooks

- Physical Science Gradeexemplar Caps 2014 eBook Subscription Services
- Physical Science Gradeexemplar Caps 2014 Budget-Friendly Options
- 6. Navigating Physical Science Gradeexemplar Caps 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Gradeexemplar Caps 2014 Compatibility with Devices
  - Physical Science Gradeexemplar Caps 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Gradeexemplar Caps 2014
  - Highlighting and Note-Taking Physical Science Gradeexemplar Caps 2014
  - Interactive Elements Physical Science Gradeexemplar Caps 2014
- 8. Staying Engaged with Physical Science Gradeexemplar Caps 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Gradeexemplar Caps 2014
- 9. Balancing eBooks and Physical Books Physical Science Gradeexemplar Caps 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Gradeexemplar Caps 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Gradeexemplar Caps 2014
  - Setting Reading Goals Physical Science Gradeexemplar Caps 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Gradeexemplar Caps 2014
  - Fact-Checking eBook Content of Physical Science Gradeexemplar Caps 2014
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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