

Flowering Leaf Disk Photosynthesis Lab

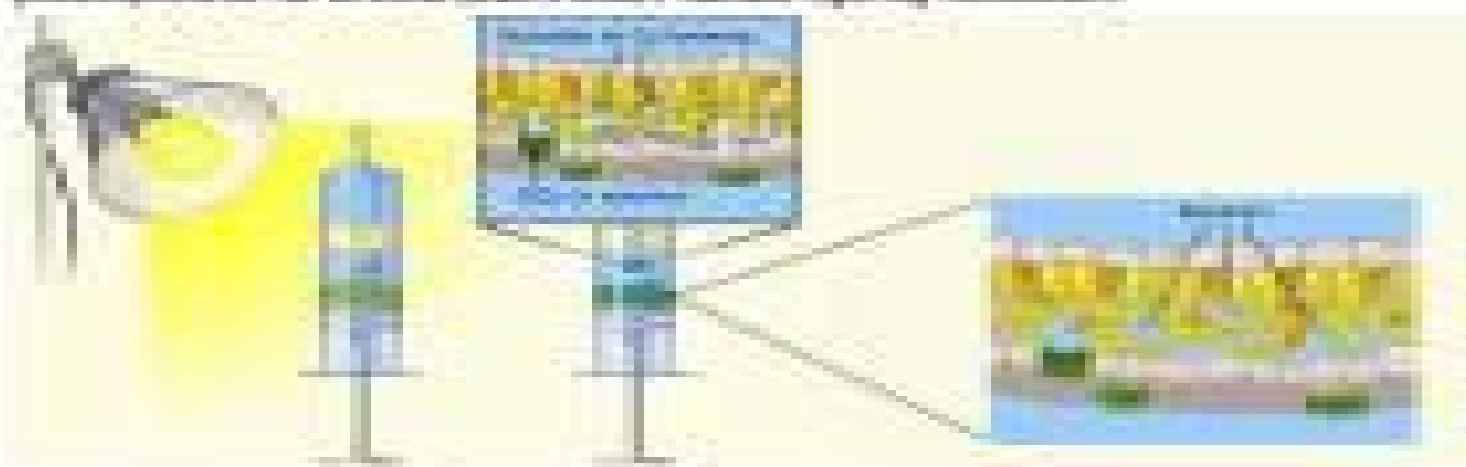
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Introduction

Light is a part of a continuous spectrum, or energy waves. Shorter wavelengths of energy have greater amounts of energy. For example, high energy ultraviolet rays, with wavelengths of approximately 1 nanometer (nm) to 400 nm, can burn living tissues that no other light sources can. Long energy waves, wavelengths of light visible to the visible part of the light spectrum, are **photosynthetic**. The visible light spectrum is 400 nm to 700 nm. 400 nm is 4 billionths of a meter. 700 nm is 7 billionths of a meter. Long visible light, with its lower-energy wavelengths, has enough energy to cause chemical changes without destroying molecules. The short, high frequency waves of gamma rays, 100,000 nm, have the most energy and break the hydrogen bonds that hold together biological molecules such as amino acid and nucleic acids, like DNA. The longer waves of most photosynthetic wavelengths (400 nm to 700 nm) do not provide enough energy and are absorbed by other water molecules in a plant.

When light is absorbed by leaf pigments such as **chlorophyll a** or **b**, absorbed visible light (photosynthesis) is used for photosynthesis. The energy is used for photosynthesis. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction.

When the rate of photosynthesis is measured, a reaction that absorbs energy and that leaf disk movement and that rate of movement. The leaf disk movement is a good measure of photosynthesis. Photosynthesis is a reaction that absorbs energy and that leaf disk movement is a good measure of photosynthesis. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction. The reaction $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ is an endothermic reaction.



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Conclusion

M Carnoy



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