

# Photoelectric Effect

When light of a suitable frequency is allowed to incident on a metal, ejection of electrons take place. This phenomenon is known as photo electric effect.

Einstein theory of photo electric effect.

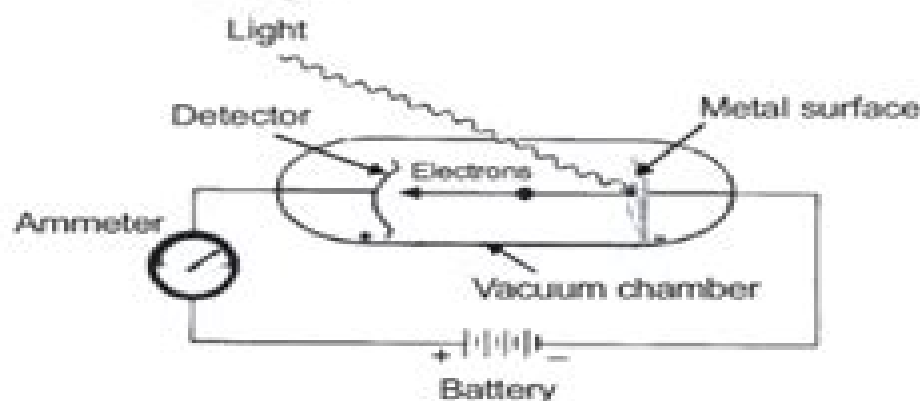


Figure : Equipment for studying the photoelectric effect

Light of a particular frequency strikes a clean metal surface inside a vacuum chamber. Electrons are ejected from the metal and are counted by a detector that measures their kinetic energy.

## Main Point

1. One photon of suitable frequency can eject one electron
2. Energy of the photon can be utilised in two ways :
  - (i) Its one part is used to just eject the electron, this part is known as work function. Work function can be calculated as

$$W = h\nu_0, \text{ where } \nu_0 \text{ is threshold frequency}$$

$$W = \frac{hc}{\lambda_0}, \lambda_0 \rightarrow \text{threshold wavelength}$$

Threshold frequency : Minimum frequency that is required to just eject electron.

- (ii) Remaining part of energy is used to provide kinetic energy.

So, Einstein equation is as given below

$$\begin{array}{c} \underline{h\nu} \\ \text{Incident} \\ \text{photon} \end{array} = \begin{array}{c} \underline{h\nu_0} \\ \text{Work} \\ \text{function} \end{array} + \begin{array}{c} \underline{KE} \\ \text{Kinetic} \\ \text{energy} \end{array}$$

# Photoelectric Effect Test Question

**Ravi Raj Dudeja**



## Photoelectric Effect Test Question:

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