

## Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds:

- 1)  $\text{NH}_4\text{Cl}$  Ammonium chloride
- 2)  $\text{Fe}(\text{NO}_3)_3$  Iron (III) Nitrate
- 3)  $\text{TiBr}_3$  Titanium (III) Bromide
- 4)  $\text{Cu}_3\text{P}$  Copper (I) Phosphide
- 5)  $\text{SnSe}_2$  Tin (IV) Selenide
- 6)  $\text{GaAs}$  Gallium Arsenide
- 7)  $\text{Pb}(\text{SO}_4)_2$  Lead (IV) Sulfate
- 8)  $\text{Be}(\text{HCO}_3)_2$  Beryllium Hydrogen Carbonate
- 9)  $\text{Mn}_2(\text{SO}_3)_3$  Manganese (II) Sulfite
- 10)  $\text{Al}(\text{CN})_3$  Aluminum Cyanide

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate  $\text{Cr}(\text{PO}_4)_2$
- 12) vanadium (IV) carbonate  $\text{V}(\text{CO}_3)_2$
- 13) tin (II) nitrite  $\text{Sn}(\text{NO}_2)_2$
- 14) cobalt (III) oxide  $\text{Co}_2\text{O}_3$
- 15) titanium (II) acetate  $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$
- 16) vanadium (V) sulfide  $\text{V}_2\text{S}_5$
- 17) chromium (III) hydroxide  $\text{Cr}(\text{OH})_3$
- 18) lithium iodide  $\text{LiI}$
- 19) lead (II) nitride  $\text{Pb}_3\text{N}_2$
- 20) silver bromide  $\text{AgBr}$

# Practice Ionic And Polyatomic Compounds Answers

**Pamela Mills,Amina El-Ashmawy**



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