

Isotopes and Isotopes Practice 105

Please define:

Proton: _____

Neutron: _____

Electron: _____

Atom: _____

Isotope: _____

Complete this statement: The atomic number of an atom is the number of _____ in that atom.

Complete the table below. The first row is done for you.



Some negative ions:

Cl^- chlorine: 17 protons and 18 electrons

O^{2-} oxygen: 8 protons and 10 electrons

N^{3-} nitrogen: 7 protons and 10 electrons

Element Symbol	Atomic Number (Z of protons)	# of neutrons	# of electrons	Atomic mass	Atomic or ionic charge	Isotope Symbol
C	6	8	5	14	+1	$^{14}_6\text{C}$
Mg			12	26		
	30	36	28			
Po		120	84			
Kr				82	0	
O		8			-2	
					+1	^6_3Li

Mg

$^{26}_{12}\text{Mg}$

$^{26}_{12}\text{Mg}$

$^{26}_{12}\text{Mg}$

Symbol for atom

Symbol for isotope

Symbol for isotope

Symbol for isotope

Practice With Isotope And Ions Key

Angie Papple Johnston

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Practice With Isotope And Ions Key:

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Factors 2 6 Rule of the Mean 2 7 Isotopic Thermometers Forest Fertilization: Theory and Practice, 1968 **Proteomics
in Practice** Reiner Westermeier, Tom Naven, Hans-Rudolf Höpker, 2008-09-08 Still the only concise practical guide to
laboratory experiments in proteomics this new edition now also covers DIGE technology and liquid chromatography while the
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