SALT ANALYSIS - SALT 4

AIM OF THE EXPERIMENT: TO ANALYSE THE GIVEN SALT FOR ACIDIC AND BASIC RADICALS.

EXPERIMENT	OBSERVATION	INFERENCE
1. COLOUR	White	Cu²+,Mn²+, Co²+, Ni²+, Fe³+ are absent
ODOUR ILAME TEST	Ammonical and vinegar like smell	NH ₄ ° or CH ₃ COO may be present .
A small amount of salt paste with conc. HCl is place on platinum wire loop and introduce into the flame.	No characteristic observation	Ca ²⁺ , Ba ²⁺ Sr ²⁺ ,Na ⁺ , K ⁺ ,Cu ²⁺ , Zn ²⁺ ,Mn ²⁺ , Pb ²⁺ are absent
FOR ACID RADICALS		·.
(i)PRELIMINARY TEST WITH DILUTE H ₂ SO ₄ Salt sample + 1- 2 ml dil. H ₂ SO ₄ . Noted the changes in cold condition and then the solution is gently made warm.	No characteristics observation ie no gas evolved	CO ₃ ²⁻ ,NO ₂ -, S ²⁻ , SO ₃ ²⁻ etc are absent
(ii)PRELIMINARY TEST WITH CONC. H ₂ SO ₄ Salt sample + about 1- 2 ml. of conc. H ₂ SO ₄ and warmed gently.	A colourless vapours evolved	May be Cl ⁻ , Br ⁻ , l ⁻ , NO ₃ ⁻ , CH ₃ COO ⁻ , C ₂ O ₄ ²⁻ are present
CONFIRMATORY TEST FOR NO ₃ = 1.COPPER CHIPS TEST Salt sample + Conc. Sulphuric acid + few copper chips	Dark brown fumes of nitrogen dioxide are evolved .	Presence of NO ₃ is confirmed .
2. RING TEST Salt sample + freshly prepared ferrous sulphate solution + conc . sulphuric acid added slowly along the sides of the test tube .	A dark brown ring is formed at the junction of the layers of the acid and the solution	Presence of NO ₃ is confirmed .

FOR BASIC RADICAL

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