

Chemistry Essay Example about Chemical Properties and Atoms Essay

Discussion Question 1 - Chemical Properties

Physical properties can be measured without changing the chemical composition of the substance and can be perceived using senses. They are observed or measured, without requiring any knowledge of the reactivity or chemical behavior of the substance, without altering its composition or its chemical nature. Some examples of physical properties of a material include texture, boiling point, melting point, hardness, texture, solubility, and temperature (Masterton & Hurley, 2009). The chemical properties of matter are what make the change of composition of matter. The exposure of any matter to a series of reagents or particular conditions can react and change its structure. Some examples of chemical properties of a material include alkalinity, pH, combustion, flammability, corrosivity, and calorific power.

A chemical reaction occurs when one or more substances, called reagents, are transformed into another or other substances having different properties, called products. When a chemical reaction takes place, the bonds between the atoms that form the reactants are broken down. The atoms rearrange themselves in a different way, resulting in the formation of new bonds, and producing one or more substances different from the initial ones (Masterton & Hurley, 2009). Chemical reactions are accompanied by a profound modification of the properties of the body or reactive bodies whereas physical reactions give rise to a very small and often partial alteration of the properties of the body (Masterton & Hurley, 2009). Chemical reactions are almost always permanent while, in general, physical reactions persist only as long as the cause that originates them. Chemical reactions are accompanied by a significant variation of energy while physical reactions are linked to a relatively small variation in energy.

A substance can be a compound or an element. Compounds are pure substances formed by two or more different elements combined in a constant proportion. All the compounds are can be divided into simpler substances, be they simpler elements or compounds. On the other hand, elements are made of the same type of atoms (for example hydrogen, H₂, Nitrogen, N₂). Elements are made up of atoms while compounds are made up of different elements (Siegfried, 2002). Some of the elements found in the body include sodium and calcium while some of the compounds present in the body include water (H₂O) and carbohydrates. Calcium strengthens the bones in the body and makes the teeth stronger. Sodium is used in stabilizing ions present in the body. Water is an essential constituent of the human body and without it, there would be no life. Carbohydrates are organic compounds composed of celluloses, sugars, and starch. They provide energy to the body and are converted to glucose, a sugar used in the body. The organic molecules that make up carbohydrates include carbon, oxygen, and hydrogen.

Discussion Question 2 - Atoms

An atom is the defining structure of the element. Elemental atoms differ from one other by the number of protons in the nucleus. Ions are a special type of atom or molecule that has a total

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