

CHEMISTRY 9TH – Short / Detailed Question Answers

➔ CHEMICAL BONDING

CHAPTER# 04

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Q.1: Define chemical bond or chemical forces.**Ans: Chemical Bond or Chemical Forces:**

All the matters in this world are composed of atoms. The attractive force which binds atoms together is called a chemical bond or chemical forces. Few elements also consist of un-bounded atoms. For instance, helium, neon, argon, xenon and krypton present in the atmosphere consist of un-bounded atoms. How various atoms are bonded together affects the properties of substances.

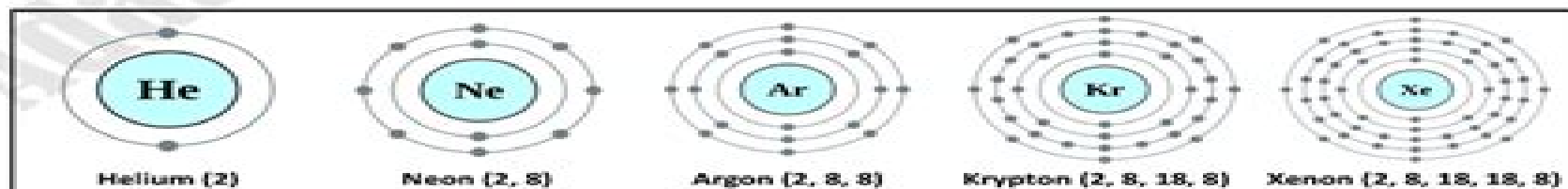
Chemical bonding is the combining of atoms to form new substances. An interaction that holds two atoms together is called a chemical bond. Atoms can lose, gain or share valence electrons to form chemical bonds.

Q.2: Why do atoms form chemical bonds?

Ans: The essential answer is that everyone in the world desires to be stable in their life. Atoms are just like that, they are also trying to become more stable, so atom tries to shares some electrons with each other to obtain the electronic configurations of noble gases.

Q.3: Describe the electronic configuration of noble gases.

Ans: Noble gases have $ns^2 np^6$ electronic configuration in the outermost shell and rarely form chemical bonds. The noble gases are helium (He), Neon (Ne), Argon (Ar), Krypton (Kr), Xenon (Xe) and Radon (Rn). These elements are sometimes called inert gases. This is because they do not participate in chemical reactions. The outer shells of three noble gas atoms are shown in the following figure.



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