

SI Chemistry

Chemical Reactions Lab

Purpose: to become more familiar with different types of chemical reactions by conducting several reactions, noting the results, and discussing observations with colleagues.

Procedure: At each of the nine stations, there are directions for you to follow-**including directions for waste disposal – FOLLOW ALL DIRECTIONS CAREFULLY!**

You should visit only six of the nine stations. You **MUST** visit stations 6, 8 and 9. You will then choose one station from stations 1 and 7, one station from stations 2 and 5 and one station from stations 3 and 4. You do not need to move through the stations in any particular order. You **MUST** wear your safety goggles at **ALL TIMES!**

At **each station**, answer the following questions in your lab notebook or on your tablet:

- a. How did you know a chemical reaction took place?
- b. Formula of reactant(s)
- c. Formula of product(s)
- d. Type of chemical reaction
- e. Write the balanced equation for the reaction (including all physical states).

Station 1: Get a strip of magnesium ribbon (about 2.5 cm long). Holding it with forceps, ignite the ribbon in a Bunsen burner flame. **Do not look directly at the burning magnesium!** **Looking directly at the burning magnesium could damage your eyes.** Hold the burning ribbon over a glass plate so that no burning magnesium lands on the lab bench.

Word equation: Magnesium reacts with oxygen to produce magnesium oxide.

Station 2: Place 10 drops of silver nitrate into a small test tube. Add 10 drops of sodium chloride to the same test tube.

Word equation: Aqueous silver nitrate reacts with aqueous sodium chloride to produce solid silver chloride and aqueous sodium nitrate.

Si Chemistry Chemical Reaction Lab Introduction

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Si Chemistry Chemical Reaction Lab Introduction:

Unveiling the Magic of Words: A Report on "**Si Chemistry Chemical Reaction Lab Introduction**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Si Chemistry Chemical Reaction Lab Introduction**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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