

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

a. Sulfur hexafluoride SF_6

b. Lithium nitride Li_3N

c. Chromium (III) carbonate $Cr_2(CO_3)_3$

d. Tin (II) chloride $TnCl_2$

e. Ammonium acetate $NH_4C_2H_3O_2$ (or CH_3COONH_4)

f. Mercury (I) chloride Hg_2Cl_2

g. Potassium bromate $KBrO_3$

h. Hydrosulfuric acid $H_2S(aq)$

i. Chloric acid $HClO_3$

j. Sulfurous acid H_2SO_3

Name each of the following compounds.

a. CuSO₄ Copper (II) sulfate

b. AlF₃ aluminum fluoride

c. HI hydrogen iodide

d. NO nitrogen monoxide

e. H₂Se(aq) hydroseconic acid

f. HNO₃ nitric acid

g. HNO₂ nitrous acid

h. NaHSO₄ sodium bisulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it takes for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives } 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20 \\ & + & \\ & & 63.62 \rightarrow 64 \end{array}$$

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John C. Rodda, Mark Robinson

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R.A. Mackay,W. Henderson,2017-12-21 This popular and comprehensive textbook provides all the basic information on inorganic chemistry that undergraduates need to know For this sixth edition the contents have undergone a complete revision to reflect progress in areas of research new and modified techniques and their applications and use of software packages Introduction to Modern Inorganic Chemistry begins by explaining the electronic structure and properties of atoms then describes the principles of bonding in diatomic and polyatomic covalent molecules the solid state and solution chemistry Further on in the book the general properties of the periodic table are studied along with specific elements and groups such as hydrogen the s elements the lanthanides the actinides the transition metals and the p block Simple and advanced examples are mixed throughout to increase the depth of students understanding This edition has a completely new layout including revised artwork case study boxes technical notes and examples All of the problems have been revised and extended and include notes to assist with approaches and solutions It is an excellent tool to help students see how inorganic chemistry applies to medicine the environment and biological topics **Solutions Guide, Introductory Chemistry, a Foundation,**

Introductory Chemistry, Basic Chemistry, Fourth Edition, Zumdahl James F. Hall,2000 [Reviews of Modern](#)

[Quantum Chemistry](#) K. D. Sen,2002 This important book collects together state of the art reviews of diverse topics covering almost all the major areas of modern quantum chemistry The current focus in the discipline of chemistry synthesis structure reactivity and dynamics is mainly on control A variety of essential computational tools at the disposal of chemists have emerged from recent studies in quantum chemistry The acceptance and application of these tools in the interfacial disciplines of the life and physical sciences continue to grow The new era of modern quantum chemistry throws up promising potentialities for further research Reviews of Modern Quantum Chemistry is a joint endeavor in which renowned scientists from leading universities and research laboratories spanning 22 countries present 59 in depth reviews Along with a personal

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Electrochemistry H R Thirsk, 2007-10-31 Specialist Periodical Reports provide systematic and detailed review coverage of progress in the major areas of chemical research Written by experts in their specialist fields the series creates a unique service for the active research chemist supplying regular critical in depth accounts of progress in particular areas of chemistry For over 80 years the Royal Society of Chemistry and its predecessor the Chemical Society have been publishing reports charting developments in chemistry which originally took the form of Annual Reports However by 1967 the whole spectrum of chemistry could no longer be contained within one volume and the series Specialist Periodical Reports was born The Annual Reports themselves still existed but were divided into two and subsequently three volumes covering Inorganic Organic and Physical Chemistry For more general coverage of the highlights in chemistry they remain a must Since that time the SPR series has altered according to the fluctuating degree of activity in various fields of chemistry Some titles have remained unchanged while others have altered their emphasis along with their titles some have been combined under a new name whereas others have had to be discontinued *Chemical News and Journal of Industrial Science*, 1865

Spectroscopic Properties of Inorganic and Organometallic Compounds N N Greenwood, 2007-10-31 Spectroscopic Properties of Inorganic and Organometallic Compounds provides a unique source of information on an important area of chemistry Divided into sections mainly according to the particular spectroscopic technique used coverage in each volume

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Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1900
Progress in Modern Hydrology John C. Rodda, Mark Robinson, 2015-08-26

Hydrology is vital to human civilisations as well as to natural ecosystems yet it has only emerged as a distinct scientific discipline during the last 50 years or so. This book reviews the development of modern hydrology primarily through the experiences of the multidisciplinary team of scientists and engineers at Wallingford near Oxford who have been at the forefront of many of the developments in UK hydrological research. These topics include:

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- Meeting the challenges of climate change
- Data handling
- Future developments in hydrology and the pressures which generate them
- Readership hydrologists in both academia and a wide range of applied fields such as civil engineering, meteorology, geography and physics as well as advanced students in earth science, environmental science and physical geography programmes worldwide.

Bibliography of Medical Reviews, 1971
Chemical Age, 1920 Lab Experiments Modern Chemistry Harold Clark Metcalfe, 1982 **International Index to Periodicals**, 1924 **MODERN PHARMACEUTICAL ANALYTICAL TECHNIQUES** Mrs. Mitali Yogesh Patil, Mr. Biprajit Paul, Mr. Mohit Chadha, Dr. Saroj Kumar Raul, In the dynamic realm of pharmaceutical sciences this project explores Modern Pharmaceutical Analytical Techniques delving into cutting edge methodologies crucial for ensuring the quality and efficacy of drugs. From spectroscopy to advanced technologies like metabolomics each chapter demystifies the application and significance of these techniques. Bridging academia and industry this work aims to be a practical guide underlining the realworld implications of these tools. Gratitude is extended to mentors, colleagues and institutions as this concise exploration seeks to serve students, researchers and professionals navigating the ever evolving landscape of pharmaceutical analysis.

Main-Group Metal Organometallics in Organic Synthesis A. McKillop, 2004-09-16 The individual chapters in this volume cover the scope and impact of main group organometallic compounds and reagents on organic synthesis during the last ten

to fifteen years In a number of chapters topics are dealt with in detail that either were not covered at all in COMC eg selenium tellurium or were given scant attention eg oxymercuration organoantimony compounds Certain topics like directed metallation and LiKOR bases have only achieved prominence in synthesis in the last ten years and are now reviewed by leading experts *Chemical News and Journal of Physical Science ,1881*

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