

First Chapter Short Questions (Fsc Part I)
Malik Xufyan
Msc Chemistry

Q4: What are ions? Under What condition are they produced?

Ans:

Ions

Ions are produced by gaining or loosing electrons .

ions are those substance which have positive or negative ion

cation are those specie which have positive charge by removal of electron

Anion are those species which have negative charges by gaining of electrons

Conditions:

Ions are produced by

- (i) By dissolving ionic compounds in water
- (ii) By X-rays
- (iii) In mass spectrometry
- (iv) By removing or adding electron in atom

Q8: Justify the following statements:

(a) 23 g of sodium and 238g of uranium have equal number of atoms in them

Ans :

23g of Na = 1 mole of Na = 6.02×10^{23} atoms of Na

238g of U = 1 mole of U = 6.02×10^{23} atoms of U.

Since equal number of gram atoms(moles) of different elements contain equal number of atoms.

Hence, 1 mole (23g) of sodium and 1 mole (238)g of uranium contain equal number of atoms ,

i , e , 6.02×10^{23} atoms.

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AN Whitehead



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Teaching First Year Chemistry Joseph Otto Frank, 1927

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appropriate to exploring such transitions The London Medical Record ,1885 The Journal of Education ,1892

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