



Urgos Grandmother Clock Manual

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 temperature change plugging in the values given in the problem you get $q_{\text{water}} = 4.18 \text{ J/g} \cdot \text{C} \times 110 \text{ g} \times 26.6 \text{ C} = 1250 \text{ J}$
 $550 \text{ J} = \Delta h_{\text{q water}} = 550 \text{ J}$

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 $14.73 \text{ kJ} = \Delta h_{\text{rxn}} = q_{\text{rxn}} / n_{\text{mcs}} \Delta t$

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