Types of Chemical Reactions Workshoot #1

HE Charmintery

For each of the following equations, identify what kind of section it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 10 2Mg + O₂ → 2 MgO Symbols
- 23 Cally + Cl2 -- CaCly + Is Single Replacement
- 2) JECOTT ANCHY -+ ANCHEN SICCI Describe Eurobecomons
- 41 C+ O₂ -+ CO₃ Symbolic
- "5) CarOffin + 2001 -- CaCly + 2000 Double Replacement
- 69 2KENOs -+ 2KEN + 3 Os Decomposition
- 7) hCubOu + 2Al → AlstSOuls + 3Cu Single Replacement
- 30 NapS + ZAgNO₂ → 2NaNO₂ + AgeS Duality Replacement
- (9) MH₂ + N₂ → 2NH₃ Symbolic
- *30) 2Na + 2000 -+ 2NaOH Hr Single Replacement

Complete each of the following equations as torough to make it the type of traction indicated. He sure to write each formula correctly.

- 111 Double replacement: NexCrO_k + PbCl_k + + 2 NaCl + PbCvO_k
- (2) Single performant: Ch + 2 Nullir -+ 2 Null + Bin-
- (3) Decomposition: MgC(3O₁)₁ → MgC(1 + 3·O₁)
- (4) Synthesis: 2 N₂ + O₂ → 2 11,O
- (5) Double replacement: 3 Ca(OHD) ↑ 2 FeCl₃ → 2 Fe(OH) ↑ 3 Ca(D₁)
- 16) Single replacement: Fe + Cu(NO₃)₂ → Fo(NO₃)₂ + Cu [Assume Fe^{2*}]
- 1.T) Decomposition: 2 Hg/O·→ 4 Hg = Cl.
- 18) Synthesis: S + O₂ -+ SO₂
- 19) Double replacement: AgNO₃ + KI → Agt + KNO₃.
- 20) Single replacement: Cu = 2 AgNO₃ -+ 2 Ag = CutNO₃; [Copper (B) is used here]

[&]quot;Remonitor "HeO" can also be thought of as "HOH!"

Science Skills Chemical Reactions Answers

Kaplan Test Prep

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