

# PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1 H Hydrogen 1.008	2 He Helium 4.0026																	
3 Li Lithium 6.941	4 Be Beryllium 9.0122																	
5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180													
11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948											
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 52.00	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.69	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.36	47 Ag Silver 107.868	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29	
55 Cs Cesium 132.91	56 Ba Barium 137.33	57-71 Lanthanide Series		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210	
87 Fr Francium 223	88 Ra Radium 226	89-103 Actinide Series		104 Rf Rutherfordium 261	105 Db Dubnium 262	106 Sg Seaborgium 266	107 Bh Bohrium 264	108 Hs Hassium 277	109 Mt Meitnerium 268	110 Ds Darmstadtium 271	111 Rg Roentgenium 272	112 Cn Copernicium 285	113 Nh Nihonium 286	114 Fl Flerovium 289	115 Uup Ununpentium 288	116 Lv Livermorium 293	117 Ts Tennessine 294	118 Og Oganesson 294

C Solid

Hg Liquid

H Gas

Rf Unknown

Metals

Nonmetals

Alkali metals

Alkaline earth metals

Lanthanides

Actinides

Transition metals

Post-transition metals

Noble gases

Atomic #

Symbol

Name

Weight

1  
H  
Hydrogen  
1.008

2  
He  
Helium  
4.0026

3  
Li  
Lithium  
6.941

4  
Be  
Beryllium  
9.0122

5  
B  
Boron  
10.81

6  
C  
Carbon  
12.011

7  
N  
Nitrogen  
14.007

8  
O  
Oxygen  
15.999

9  
F  
Fluorine  
18.998

10  
Ne  
Neon  
20.180

11  
Na  
Sodium  
22.990

12  
Mg  
Magnesium  
24.305

13  
Al  
Aluminum  
26.982

14  
Si  
Silicon  
28.086

15  
P  
Phosphorus  
30.974

16  
S  
Sulfur  
32.06

17  
Cl  
Chlorine  
35.45

18  
Ar  
Argon  
39.948

19  
K  
Potassium  
39.098

20  
Ca  
Calcium  
40.078

21  
Sc  
Scandium  
44.956

22  
Ti  
Titanium  
47.88

23  
V  
Vanadium  
50.942

24  
Cr  
Chromium  
52.00

25  
Mn  
Manganese  
54.938

26  
Fe  
Iron  
55.845

27  
Co  
Cobalt  
58.933

28  
Ni  
Nickel  
58.69

29  
Cu  
Copper  
63.546

30  
Zn  
Zinc  
65.38

31  
Ga  
Gallium  
69.723

32  
Ge  
Germanium  
72.63

33  
As  
Arsenic  
74.922

34  
Se  
Selenium  
78.96

35  
Br  
Bromine  
79.904

36  
Kr  
Krypton  
83.80

37  
Rb  
Rubidium  
85.468

38  
Sr  
Strontium  
87.62

39  
Y  
Yttrium  
88.906

40  
Zr  
Zirconium  
91.224

41  
Nb  
Niobium  
92.906

42  
Mo  
Molybdenum  
95.94

43  
Tc  
Technetium  
98

44  
Ru  
Ruthenium  
101.07

45  
Rh  
Rhodium  
102.91

46  
Pd  
Palladium  
106.36

47  
Ag  
Silver  
107.868

48  
Cd  
Cadmium  
112.41

49  
In  
Indium  
114.82

50  
Sn  
Tin  
118.71

51  
Sb  
Antimony  
121.76

52  
Te  
Tellurium  
127.6

53  
I  
Iodine  
126.905

54  
Xe  
Xenon  
131.29

55  
Cs  
Cesium  
132.91

56  
Ba  
Barium  
137.33

57-71  
Lanthanide Series

72  
Hf  
Hafnium  
178.49

73  
Ta  
Tantalum  
180.948

74  
W  
Tungsten  
183.84

75  
Re  
Rhenium  
186.207

76  
Os  
Osmium  
190.23

77  
Ir  
Iridium  
192.22

78  
Pt  
Platinum  
195.084

79  
Au  
Gold  
196.967

80  
Hg  
Mercury  
200.59

81  
Tl  
Thallium  
204.38

82  
Pb  
Lead  
207.2

83  
Bi  
Bismuth  
208.98

84  
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209

85  
At  
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89-103  
Actinide Series

104  
Rf  
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261

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262

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Sg  
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107  
Bh  
Bohrium  
264

108  
Hs  
Hassium  
277

109  
Mt  
Meitnerium  
268

110  
Ds  
Darmstadtium  
271

111  
Rg  
Roentgenium  
272

112  
Cn  
Copernicium  
285

113  
Nh  
Nihonium  
286

114  
Fl  
Flerovium  
289

115  
Uup  
Ununpentium  
288

116  
Lv  
Livermorium  
293

117  
Ts  
Tennessine  
294

118  
Og  
Oganesson  
294

270

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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## Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol <sup>1</sup>H), has a single proton and zero neutrons.

## Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

## Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

# Periodic Table Basic Speedy Study Guides

**Iris Stovall, Steven S. Zumdahl**



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An advanced

periodic table of elements displays not only the elements but the ions that form each element A pamphlet with such a visual aid would greatly benefit chemistry students Any student taking chemistry will need to learn the elements A pamphlet would be concise and break the information down simply making it easier to understand and remember It allows students to simply focus on the main point rather than taking in information that they may or may not need      **Organic Chemistry**

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**Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry** Iris Stovall,Steven S. Zumdahl,1996      *TEExES Core Subjects 4-8 (211) Book + Online, 2nd Ed.* Ann M.L. Cavallo,Karen Allmond,Mary D. Curtis,Marci Smith Deal,Christina Gawlik,Candace Joswick,Melissa Hulings,Kathleen C. Tice,2023-01-03

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