

Periodic Table: Virtual Lab Document



Part 1:

Follow the instructions in the Periodic Table Virtual Lab Lesson to color the periodic table.

Click on the image below -> edit -> drop down arrow beside the line -> select tool -> change colors.

H																		He
Li	Be											B	C	N	O	F	Ne	
Na	Mg											Al	Si	P	S	Cl	Ar	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Uun	Uuu	Uub							
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

Part 2: Go to chemicalelements.com.

1. Click on Alkali Metals (left bar) to answer the following questions:

- What is the group number? [1]
- Are these metals reactive? [Yes, they are highly reactive]
- Do these metals occur freely in nature? [No, they don't occur freely in nature, but they are found in various minerals.]
- How many electrons are in their outer shell? [1]
- What are the three characteristics of ALL metals? [Shiny luster, good conductors of heat and electricity, malleable and ductile.]
- Are these metals soft or hard? [soft]
- Name the two most reactive elements in this group? [Potassium and Sodium]
- What happens when they are exposed to water? [They react vigorously with water, producing hydrogen gas and an alkaline solution. The reaction can be violent and can even ignite the hydrogen gas produced.]

Periodic Table Lab Answer Key

Marcia L. Tate

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Periodic Table Lab Answer Key:

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study. Features: Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule. Workflow: Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade. About the Author: DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and the Sable Institute of Environmental Studies.

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of *Hands On General Science Activities with Real Life Applications*, Pam Walker and Elaine Wood have completely revised and updated their must-have resource for science teachers of grades 5-12. The book offers a dynamic collection of classroom-ready lessons, projects, and lab activities that encourage students to integrate basic science concepts and skills into everyday life.

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Handbook of Formative Assessment Heidi Andrade, Gregory J. Cizek, 2010-04-02 The Handbook of Formative Assessment

comprehensively profiles this burgeoning field of study Written by leading international scholars and practitioners each chapter discusses key issues in formative assessment policy and practice

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philosophers and scientists focusing on the debate in science between those who believe that science is above criticism and those who do not

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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TABLE

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Table of Contents **Periodic Table Lab Answer Key**

1. Understanding the eBook Periodic Table Lab Answer Key
 - The Rise of Digital Reading Periodic Table Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Lab Answer Key
 - Personalized Recommendations
 - Periodic Table Lab Answer Key User Reviews and Ratings
 - Periodic Table Lab Answer Key and Bestseller Lists

5. Accessing Periodic Table Lab Answer Key Free and Paid eBooks
 - Periodic Table Lab Answer Key Public Domain eBooks
 - Periodic Table Lab Answer Key eBook Subscription Services
 - Periodic Table Lab Answer Key Budget-Friendly Options
6. Navigating Periodic Table Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Lab Answer Key Compatibility with Devices
 - Periodic Table Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Lab Answer Key
 - Highlighting and Note-Taking Periodic Table Lab Answer Key
 - Interactive Elements Periodic Table Lab Answer Key
8. Staying Engaged with Periodic Table Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Lab Answer Key
9. Balancing eBooks and Physical Books Periodic Table Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Lab Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Lab Answer Key
 - Setting Reading Goals Periodic Table Lab Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Lab Answer Key
 - Fact-Checking eBook Content of Periodic Table Lab Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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