

5.4

THE PERIODIC TABLE: ORGANIZING THE ELEMENTS

SECTION REVIEW

Objectives

- Describe the layout of the periodic table
- Identify the position of groups, periods, and the transition metals in the periodic table

Key Terms

- | | | |
|---------------------------|-------------------------|---------------|
| • periodic table | • metals | • non-metals |
| • periods | • alkali metals | • halogens |
| • periodic law | • alkaline earth metals | • noble gases |
| • group | • transition metals | • metalloids |
| • representative elements | • lanthanide elements | |

Part A Completion

Use this worksheet to complete each blank. You are allowed to refer to the periodic table and other resources as you work. Each blank can be completed with a word, a term, a phrase, or a sentence.

- The periodic table organizes the elements into vertical _____ 1. _____ and horizontal _____ 2. _____ in order of increasing _____ 3. _____. The table is constructed so that elements that have similar chemical properties are in the same _____ 4. _____. The elements in Groups 1A through 7A are called the _____ 5. _____. The _____ 6. _____ make up Group 8. The elements in Groups 9A and 10A are interrupted in periods 4 and 5 by the _____ 7. _____ and in the periods 6 and 7 by the _____ 8. _____. The Group 1A elements are called the _____ 9. _____ and the Group 2A elements are called the _____ 10. _____. The non-metals of _____ 11. _____ are called the halogens. Elements with properties that are intermediate between those of metals and non-metals are called _____ 12. _____.

Periodic Table Section Review Answer Key

Dr. Dennis Englin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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