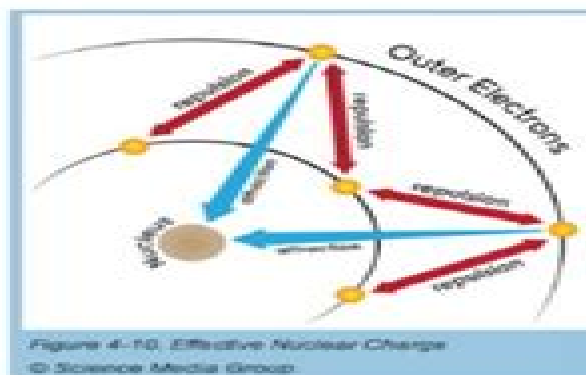


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Day 36 Coulombic Attraction and Periodic Trends POGIL

What variables will affect the force of attraction between charged particles?



Look at the image to the left, what particles are attracted to each other and what particles repel each other?

What causes them to be attracted or repelled? Please explain.

Coulombic attraction or electrostatic attraction is the attraction between oppositely charged particles. For example, the protons in the nucleus of an atom have attraction for the electrons surrounding the nucleus. This is because the protons are positive and the electrons are negative. The attractive force can be weak or strong. In this activity, you will explore the strength of attraction between protons and electrons in various atomic structures.

Model 1 – Distance and Attractive Force

		Force of Attraction (Newtons)
A		2.30×10^{-8}
B		0.58×10^{-8}
C		0.26×10^{-8}

1. What subatomic particles do these symbols represent in Model 1?



2. Would you expect to observe attraction or repulsion between the subatomic particles in Model 1?



3. Consider the data in Model 1.

a. What are the independent and dependent variables in the data? (Independent is what is being changed and dependent is what is being measured or has changed due to those changes you

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