# CHAPTER 6 STUDY GUIDE

# The Periodic Table and Periodic Law

#### Section 6.1 Development of the Modern Periodic Table

In your textbook, reads about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

octaves	atomic mass	atomia-number	The state of the s
colomerus	<u>properties -</u>	Henry Messley	
protons	periodic law	Desira Mandalosv	accented

1 2	3	4	5	6	7	
н	Li	G	Bo	C	N	0
8	9	10	TI	12	13	1.5
E	Na	Mg	: AJ	Si	P	S

The first periodic table is mostly credited to (5) Mendelety In his table, the elements were arranged according to increasing (6) Office Mass. One important result of this table was that the existence and properties of undiscovered

(7) Elements could be predicted.

The element in the modern periodic table are arranged according to increasing

(8) Offmic number, as a result of the work of (9) Moseloy. This arrangement is based on number of (10) Profos in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) Periodic last. which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) Properties.

# **Periodic Law Study Guide Answers**

J Elliott

**Periodic Law Study Guide Answers:** 

The book delves into Periodic Law Study Guide Answers. Periodic Law Study Guide Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and indepth insights into Periodic Law Study Guide Answers, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Periodic Law Study Guide Answers
  - Chapter 2: Essential Elements of Periodic Law Study Guide Answers
  - Chapter 3: Periodic Law Study Guide Answers in Everyday Life
  - Chapter 4: Periodic Law Study Guide Answers in Specific Contexts
  - ∘ Chapter 5: Conclusion
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