

Chemistry: Periodic Table Unit Study Guide

Objectives:

1. Define the following terms:
 - Element
 - Atom
 - Molecule
 - Compound
 - Mixture
2. State the Periodic Law.
3. An element is defined as a substance that cannot be broken down into simpler substances by chemical means.
4. Define the following terms:
 - Atomic number
 - Mass number
 - Isotope
5. Define the following terms:
 - Electron
 - Proton
 - Neutron
6. Define the following terms:
 - Electron shell
 - Valence electrons
7. Define the following terms:
 - Ion
 - Cation
 - Anion
8. Define the following terms:
 - Electronegativity
 - Ionization energy
9. Define the following terms:
 - Atomic radius
 - Atomic mass
10. Define the following terms:
 - Periodic table
 - Period
 - Group
 - Block
11. Define the following terms:
 - Metals
 - Nonmetals
 - Metalloids
12. Define the following terms:
 - Alkali metals
 - Alkaline earth metals
 - Transition metals
 - Inner transition metals
 - Halogens
 - Noble gases

Name: _____ Date: _____ Period: _____

Periodic Table Unit Study Guide

Objectives and Overview of the Periodic Table

1. Define the following terms:

- Element
- Atom
- Molecule
- Compound
- Mixture

2. State the Periodic Law.

3. An element is defined as a substance that cannot be broken down into simpler substances by chemical means.

4. Define the following terms:

- Atomic number
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5. Define the following terms:

- Electron
- Proton
- Neutron

6. Define the following terms:

- Electron shell
- Valence electrons

7. Define the following terms:

- Ion
- Cation
- Anion

8. Define the following terms:

- Electronegativity
- Ionization energy

9. Define the following terms:

- Atomic radius
- Atomic mass

10. Define the following terms:

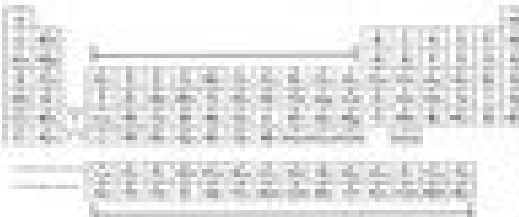
- Periodic table
- Period
- Group
- Block

11. Define the following terms:

- Metals
- Nonmetals
- Metalloids

12. Define the following terms:

- Alkali metals
- Alkaline earth metals
- Transition metals
- Inner transition metals
- Halogens
- Noble gases



s-block metals


p-block nonmetals

d-block transition metals

f-block inner transition metals

13. Match the periodic table blocks to their corresponding elements.

14. Fill in the following table:

| Element | Astatine | Ca | $1s^2 2s^2 2p^6 3s^2$ |  |
|---------------------|----------|----|-----------------------|---|
| Number of protons | | | | |
| Number of electrons | | | | |
| Atomic number | | | | |
| Mass number | | | | |

Periodic Table And Chemistry Study Guide Answers

Arshad Iqbal



Periodic Table And Chemistry Study Guide Answers:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011

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Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it

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