

Periodic Trends Worksheet

Using the periodic table and your knowledge of periodic trends, answer the following questions:

1 Circle the best answer that has the:

- | | | | | | | |
|---|----------------------------|----|----|----|----|----|
| a | smallest electron affinity | N | P | As | Sb | Bi |
| b | largest electronegativity | Ga | Ge | As | Sc | |
| c | largest ionization energy | O | S | Sc | Te | Po |
| d | lowest ionization energy | O | S | Se | Te | Po |
| e | smallest atomic radius | F | Cl | Br | I | At |
| f | largest atomic radius | Li | Na | K | Rb | Cs |

2 Low ionization energy is a characteristic of a _____

3 What does atomic radius measure? _____

4 Describe the change in the atomic size across a period and down a group.

5 Which element has the largest atomic radius? _____

6 Explain the main difference between metals and nonmetals.

7 Determine if the listed pairs of elements have similar or different properties.

- | | | | |
|---|----------------------|---|--------------------|
| a | Potassium & Rubidium | b | Calcium & Barium |
| c | Sodium & Chlorine | d | Helium & Krypton |
| e | Phosphorus & Oxygen | f | Lithium & Fluorine |

8 Which element has the largest electronegativity? _____

9 Which group (name) would have the lowest ionization energies?

10 Which group (name) has the highest ionization energies?

Periodic Trends And Ions Review Answer Sheet

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regular basis *Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *Energetics of Organometallic Species* José A. Martinho Simões, 2012-12-06 An overview of modern organometallic thermochemistry made by some of the most active scientists in the area is offered in this book The contents correspond to the seventeen lectures delivered at the NATO ASI Energetics of Organometallic Species Curia Portugal September 1991 plus three other invited contributions from participants of that summer school These papers reflect a variety of research interests and discuss results obtained with several techniques It is therefore considered appropriate to add a few preliminary words attempting to bring some unity out of that diversity In the first three chapters results obtained by classical calorimetric methods are described Modern organometallic thermochemistry started in Manchester with Henry Skinner and his pioneering work is briefly surveyed in the first chapter The historical perspective is followed by a discussion of a very actual issue the trends of stepwise bond dissociation enthalpies Geoff Pilcher another Manchester thermochemist makes in chapter 2 a comprehensive and authoritative survey of problems found in the most classical of thermochemical techniques combustion calorimetry applied to organometallic compounds Finally results from another classical technique reaction solution calorimetry are reviewed in the third chapter by Tobin Marks and coworkers More than anybody else Tobin Marks has used thermochemical values to define synthetic strategies for organometallic compounds thus indicating an application of thermochemical data of which too little use has been made so far Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis **Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation *Merrill*

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