

Photoelectric Effect

When light of a suitable frequency is allowed to incident on a metal, ejection of electrons take place. This phenomenon is known as photo electric effect.

Einstein theory of photo electric effect.

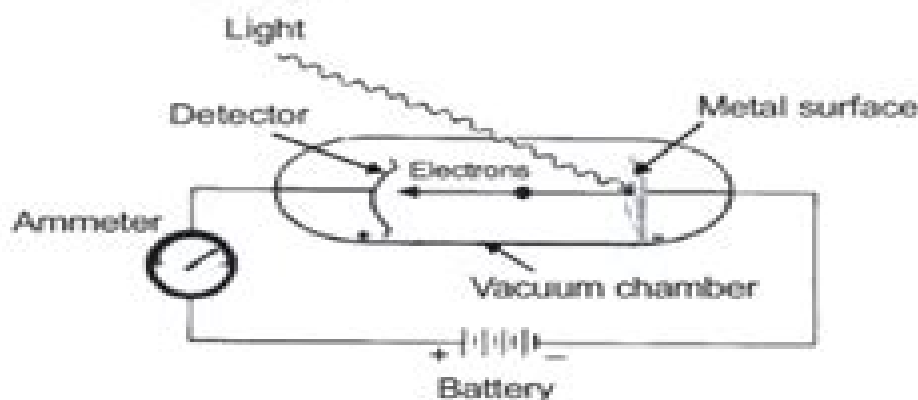


Figure : Equipment for studying the photoelectric effect

Light of a particular frequency strikes a clean metal surface inside a vacuum chamber. Electrons are ejected from the metal and are counted by a detector that measures their kinetic energy.

Main Point

1. One photon of suitable frequency can eject one electron
2. Energy of the photon can be utilised in two ways :
 - (i) Its one part is used to just eject the electron, this part is known as work function. Work function can be calculated as

$$W = h\nu_0, \text{ where } \nu_0 \text{ is threshold frequency}$$

$$W = \frac{hc}{\lambda_0}, \lambda_0 \rightarrow \text{threshold wavelength}$$

Threshold frequency : Minimum frequency that is required to just eject electron.

- (ii) Remaining part of energy is used to provide kinetic energy.

So, Einstein equation is as given below

$$\underbrace{h\nu}_{\text{Incident photon}} = \underbrace{h\nu_0}_{\text{Work function}} + \underbrace{KE}_{\text{Kinetic energy}}$$

Photoelectric Effect Test Question

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Table of Contents Photoelectric Effect Test Question

1. Understanding the eBook Photoelectric Effect Test Question
 - The Rise of Digital Reading Photoelectric Effect Test Question
 - Advantages of eBooks Over Traditional Books
2. Identifying Photoelectric Effect Test Question
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Photoelectric Effect Test Question
 - User-Friendly Interface
4. Exploring eBook Recommendations from Photoelectric Effect Test Question
 - Personalized Recommendations
 - Photoelectric Effect Test Question User Reviews and Ratings
 - Photoelectric Effect Test Question and Bestseller Lists
5. Accessing Photoelectric Effect Test Question Free and Paid eBooks
 - Photoelectric Effect Test Question Public Domain eBooks
 - Photoelectric Effect Test Question eBook Subscription Services
 - Photoelectric Effect Test Question Budget-Friendly Options
6. Navigating Photoelectric Effect Test Question eBook Formats
 - ePub, PDF, MOBI, and More
 - Photoelectric Effect Test Question Compatibility with Devices
 - Photoelectric Effect Test Question Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Photoelectric Effect Test Question
 - Highlighting and Note-Taking Photoelectric Effect Test Question
 - Interactive Elements Photoelectric Effect Test Question
8. Staying Engaged with Photoelectric Effect Test Question

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Photoelectric Effect Test Question
- 9. Balancing eBooks and Physical Books Photoelectric Effect Test Question
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Photoelectric Effect Test Question
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Photoelectric Effect Test Question
 - Setting Reading Goals Photoelectric Effect Test Question
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Photoelectric Effect Test Question
 - Fact-Checking eBook Content of Photoelectric Effect Test Question
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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