

- A. Physical property**—any characteristic of matter that can be observed without changing the identity of the material. Scientists call a characteristic of matter that you can detect with your senses a **Physical property**.

*A description of an object based on what you observe with your senses includes physical properties.

1. Physical properties include **color, shape, smell, taste, mass, volume, and density**.

2. **Density**— is a **physical property** that determines amount of mass in a given volume; a golf ball would have greater density than a table-tennis ball; the formula for density is:

$$\text{density (g/cm}^3\text{)} = \text{mass(g)}/\text{volume (cm}^3\text{)}$$

*Density is determined by dividing the **mass** of a sample by the **volume** of the sample

***Mass** is the amount of matter in an object.

*If the volume of an object is 700 cm³ and the mass is 5,000 g. What is its density?

$$*D=M/V \quad 5,000/700= 7.1428571 \quad 7.14 \text{ g/cm}^3$$

3. Density can be used to identify unknown substances; a person could determine the composition of a piece of jewelry by comparing its density with the density of a known substance such as silver or gold.

4. **State of matter**—whether a substance is a **solid, liquid, or a gas** at a particular temperature and pressure.

*Three states of matter are **solid, liquid, a gas**

5. Some physical properties are size-dependent (length, width, height, volume, and mass)

while others are size-independent (density, color, state).

* Some properties change when the size of an object changes. These properties are called **size-dependent properties**.

*Volume is **NOT** a size-independent property. Volume is an example of a size-dependent property.

**Density is an example of a size-independent property

B. Matter can be classified as an acid or a base.

1. Acids tend to have a sharp smell and a sour taste (but you should NEVER taste anything in a laboratory) with a pH between 0 and 7; citrus fruits contain acids.

2. Bases feel slippery and taste bitter and have a pH of between 7 and 14; a bar of soap is a base.

3. Water, with a pH of exactly 7, is neither an acid nor a base; it is neutral.

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