A. Physical property—any characteristic of matter that can be observed without changing the identity of the material. Scientists call a characteristic of matter that you can detect with your senses a Physical property.

> *A description of an object based on what you observe with your senses includes physical properties.

1. Physical properties include color, shape, smell, taste, mass, volume, and density.

Density— is a physical property that determines amount of mass in a given volume; a golf ball
would have greater density than a table-tennis ball; the formula for density is:

density (g/cm3) = mass(g)/volume (cm3)

*Density is determined by dividing the mass of a sample by the volume of the sample

*Mass is the amount of matter in an object

*If the volume of an object is 700 cm3 and the mass is 5,000 g. What is its density?

*D=M/V 5,000/700= 7.1428571 7.14 g/cm3

Density can be used to identify unknown substances; a person could determine the composition of a
piece of jewelry by comparing its density with the density of a known substance such as silver or gold.

4. State of matter—whether a substance is a solid, liquid, or a gas at a particular temperature and pressure.

*Three states of matter are solid, liquid, a gas

Some physical properties are size-dependent (length, width, height, volume, and mass) while others are size-independent (density, color, state).

 Some properties change when the size of an object changes. These properties are called sizedependent properties.

*Volume is **NOT** a size-independent property. Volume is an example of a <u>size-dependent</u> property

- **Density is an example of a size-independent property
- B. Matter can be classified as an acid or a base.
 - Acids tend to have a sharp smell and a sour taste (but you should NEVER taste anything in a laboratory) with a pH between 0 and 7; citrus fruits contain acids.
 - 2. Bases feel slippery and taste bitter and have a pH of between 7 and 14; a bar of soap is a base.
 - 3. Water, with a pH of exactly 7, is neither an acid nor a base; it is neutral.

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M Lipman

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