

Chapter 5 The Periodic Table

Section 5.2 The Modern Periodic Table

(pages 130–138)

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–133. As you read, write answers to your questions. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Questions About the Periodic Table	
Question	Answer
Students might ask: What does atomic mass mean?	Atomic mass is a value that depends on the distribution of an element's isotopes in nature and the masses of those isotopes.
Why are two columns of elements placed before the main body of the table?	Two kinds of elements are placed before the table to make the table more compact.

Other possible questions: Why are there two numbering systems for the columns? Why is period 7 incomplete?

The Periodic Law (pages 131–133)

- Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
_____ **true** _____
- Explain why the number of elements per period varies. _____
_____ **The number of elements per period varies because the number of available orbitals increases from energy level to energy level.** _____
- Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the _____
_____ **periodic law** _____

Atomic Mass (page 134)

- Label the four types of information supplied for chlorine in the diagram.

a.	17
b.	Cl
c.	Chlorine
d.	35.453

- Atomic number**
- Element symbol**
- Element name**
- Atomic mass**

Section 5 2 The Modern Periodic Table Reading Guide

Disha Experts

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Disha Experts' text.

Section 5 2 The Modern Periodic Table Reading Guide:

Study Guide for Chemical Principles Thomas Elliott Taylor, 1979 A New Approach to ICSE Chemistry for Class IX (A.Y. 2023-24) Onward V. K. Sally, 2023-05-20 The book A New Approach to I C S E Chemistry for Class IX has been written in accordance with the latest syllabus prescribed by the Council for Indian School Certificate Examination New Delhi The book is divided into five distinct parts 1 The first part constitutes chapters on The Languages of chemistry Chemical Changes and Reactions and Water to give an idea to the students about the diversity of matter 2 The second part constitutes chapters on Atomic Structure and Chemical Bonding and The Periodic Table This part will help the students to understand unity in diversity of matter It further explains the rules required for reaction between various forms of matter 3 The third part constitutes chapter on Study of the First Element Hydrogen and Study of Gas Laws This will familiarise the students with element hydrogen and their chemical properties It further explains the behaviour of gases under changes of temperature and pressure and also provides explanation in terms of molecular motion 4 The fourth part deals with Atmospheric Pollution 5 The fifth part deals with Practical Chemistry The salient features of the book Large number of chemical reactions are described with experimental observations Important points have been highlighted for location of precise answers All definitions and other important points are highlighted Diagrams are made simple and more Informative In the chapter on the language of chemistry a new method has been introduced for balancing chemical equations This method Is based on atomic numbers which Is fundamental property of the elements A chapter on Practical Chemistry is added to help students for laboratory work Questions have been added chapter wise under the heading Question Bank for the benefit of students The solution can be accessed through QR Code given at the end of each chapter Periodic Table showing mass number atomic number of various elements along with list of actual names of the elements is provided A Specimen Paper Solved has also been included through QR Code for the benefit of students Annual Examination Paper 2019 Solved has been given through OR Code Model Test Papers 1 to 5 Unsolved have been given through OR Code We hope this book will prove very useful to the students and teachers Suggestions and constructive criticism for the further improvement of the book would be gratefully acknowledged Author *Guide to RRB Junior Engineer Stage II Exam - Physics, Chemistry, General Awareness, Basics of Computers, Environment & Pollution Control 2nd Edition* Disha Experts, The book Guide to RRB Junior Engineer Stage II Online Exam has 4 sections common to all streams General Awareness Physics Chemistry Basics of Computers and Applications Basics of Environment and Pollution Control Each section is further divided into chapters which contains theory explaining the concepts involved followed by MCQ exercises The book provides the past 2014 2015 2019 Solved Questions The detailed solutions to all the questions are provided at the end of each chapter **Study Guide Central Hindu School Entrance Exam 2022 For Class 11** Arihant Experts, 2022-03-05 1 Central Hindu School Entrance Test is a complete test guide 2 Covers entire syllabus for class 11th 3 Topically divided into 5 sections to provide

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