

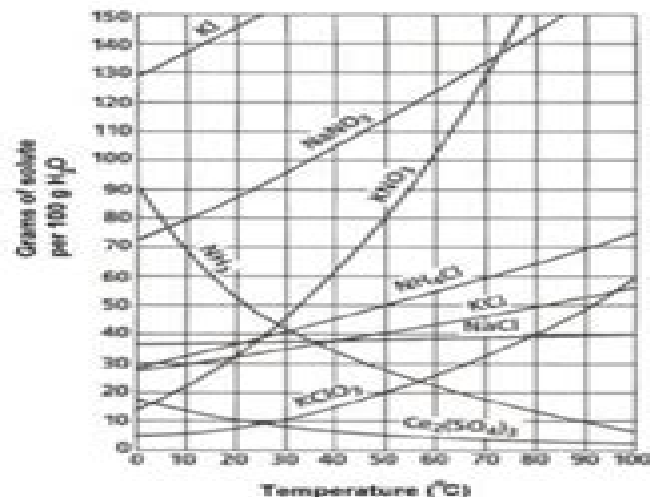
## Solubility worksheet

Write the correct term (or terms) in the space provided.

1. As temperature increases, the solubility of gases in liquids generally \_\_\_\_\_.
2. When a solute dissolves and recrystallizes at the same rate, the solution is at \_\_\_\_\_.
3. A substance that does not dissolve in a polar solvent is probably \_\_\_\_\_.
4. To carry an electric current, a solution must contain \_\_\_\_\_.
5. A solution that contains more dissolved solute than a saturated solution contains under the same conditions is called a(n) \_\_\_\_\_.
6. The solubility of  $\text{CO}_2(\text{g})$  would \_\_\_\_\_ with increasing solvent temperature.

Based on the graph answer the following questions:

1. Which compound is LEAST soluble at  $10^\circ\text{C}$ ? \_\_\_\_\_
2. How many grams of  $\text{KCl}$  can be dissolved in 100g of water at  $80^\circ\text{C}$ ? \_\_\_\_\_
3. How many grams of  $\text{NaCl}$  can be dissolved in 100g of water at  $90^\circ\text{C}$ ? \_\_\_\_\_
4. At  $40^\circ\text{C}$ , how much  $\text{KNO}_3$  can be dissolved in 100g of water? \_\_\_\_\_
5. A saturated solution of  $\text{KClO}_3$  is formed from one hundred grams of water. If the saturated solution is cooled from  $90^\circ\text{C}$  to  $50^\circ\text{C}$ , how many grams of precipitate are formed? \_\_\_\_\_



What are 3 factors that affect solubility?

- a) \_\_\_\_\_
- b) \_\_\_\_\_
- c) \_\_\_\_\_

# Section 8 2 Solubility And Concentration Worksheet Answers

**JL Elias**



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