Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)

Name of Reaction Type

- 1. $2 \underbrace{Na}_{B} + \underbrace{Cl_{2}}_{A} \Rightarrow 2 \underbrace{NaCl}_{A}_{B}$ (A & B combine) Ex: synthesis
- Pb + FeSO₄ → PbSO₄ + Fe
- P₄ + 3 O₂ → 2 P₂O₃
- 2 NO₂ → 2 O₂ + N₂
- Na₃PO₄ + 3 KOH → 3 NaOH + K₃PO₄
- C₃H₆O + 4 O₂ → 3 CO₂ + 3 H₂O
- MgCl₂ + Li₂CO₃ → MgCO₃ + 2 LiCl
- C₆H₁₂ + 9 O₂ → 6 CO₂ + 6 H₂O
- CaCO₃ → CaO + CO₂
- 2 AgNO₃ + Cu → Cu(NO₃)₂ + 2 Ag
- 2H₂ + O₂ → 2H₂O

Section 2neutralization Reactions Answers

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

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