#### Module 3: Elements and the Periodic Table

#### Directions

Click on the following website: www.ptable.com

You will interact with this periodic table to find the answers to the questions below.

#### Part 1; Atomic Numbers and Atomic Masses

Complete the following table, by finding the atomic numbers and atomic masses for each element:

Symbol	P	S	CI	Ar	K	Ca
Atomic number	15	14	17	18	19	20
Atomic Mass	30.974	28.085 5	35.453	39.948	39.0983	40.078

- a. Using the data in the table above, is the periodic table organized by increasing atomic mass? Explain. Yes, because higher atomic numbers mean more subatomic particles which increase mass
- b. What property is the periodic table organized by? Atomic number

#### Part 2: Groups and Periods:

Continue to work in this site: www.ptable.com

Make sure the "Wikipedia" tab is selected in order to answer the following questions:

	Write your answers here
The vertical columns on the periodic table are called groups, how many groups are there on the periodic table?	16
2. The horizontal rows on the periodic table are called periods, how many periods are there on the periodic table?	7
Which element is located in group 10 and period 5	Palladium
Which element is located in group 15 and period 4	Arsenic
5. Which element is located in group 2 and period 3	Magnesium
Which element is located in group 18 and period 6	Radon
7. Which element is located in group 1 and period 7	Francium
3. Are most of the elements on the periodic table classified as a metal or a non-metal?	Metals
Are the non-metal elements located on the left or right side of the periodic table?	Right

# **Section 3 The Periodic Table Answers**

**Paul Foglino** 

#### **Section 3 The Periodic Table Answers:**

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