Chemistry Finals Study Guide

Chapter 1 & 2 - Review of Basic Concepts

- Describe the 3 states of matter. The three states of matter are solid, liquid, and gas.
 - Solid: When atoms in a material are moving slow that they stick together and barely move, creating a hard object classified as a solid
 - Liquid: When atoms are moving slower than a gas state but faster than the solid state so they flow and are able to take the shape of their container. They have more energy than solid state atoms.
 - Gas: When atoms are moving faster than the other two states of matter (solid & liquid) that they are no longer pulled towards each other and roam freely without boundaries if not kept in a container.
- Describe the difference between elements and compounds, mixtures.
 - Elements: materials that are only made out of the same type of atoms and cannot be changed into simpler substances by a chemical change (ex: O₂)
 - Compounds: materials that are a mix of two or more element in a fixed proportion
 - Mixtures: a mixture of 2 or more pure substances.
 - Heterogeneous Mixture: can often be separated by filtration
 - Homogenous Mixture: can be separated by distillation, crystallization, and chromatography
- Convert between Celsius and Kelvin.
 - Celsius to Kelvin: C = (°K 273.15)
 - Kelvin to Celsius: K = (°C + 273.15)
- Round measured and calculated values to the correct number of significant digits.
 - Significant Digit Rules: [CHEM] Final 1 Steps & Processes
- Use dimensional analysis to solve multi step equations, showing how units cancel
 [CHEM] Final 1 Steps & Processes
- What is the difference between precision and accuracy?
 - Precision: Acquiring the same result again and again
 - Accuracy: A result that is close to the accepted value/actual answer

Section Chem Study Guide Answers

Neil E. Schore

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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems **Organic Chemistry Study Guide with Solutions** Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **Chemical Principles Study Guide/Solutions** Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

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